

Surface-Atmosphere Exchange of Ammonia in a Non-fertilized Grassland and its Implications for PM_{2.5}

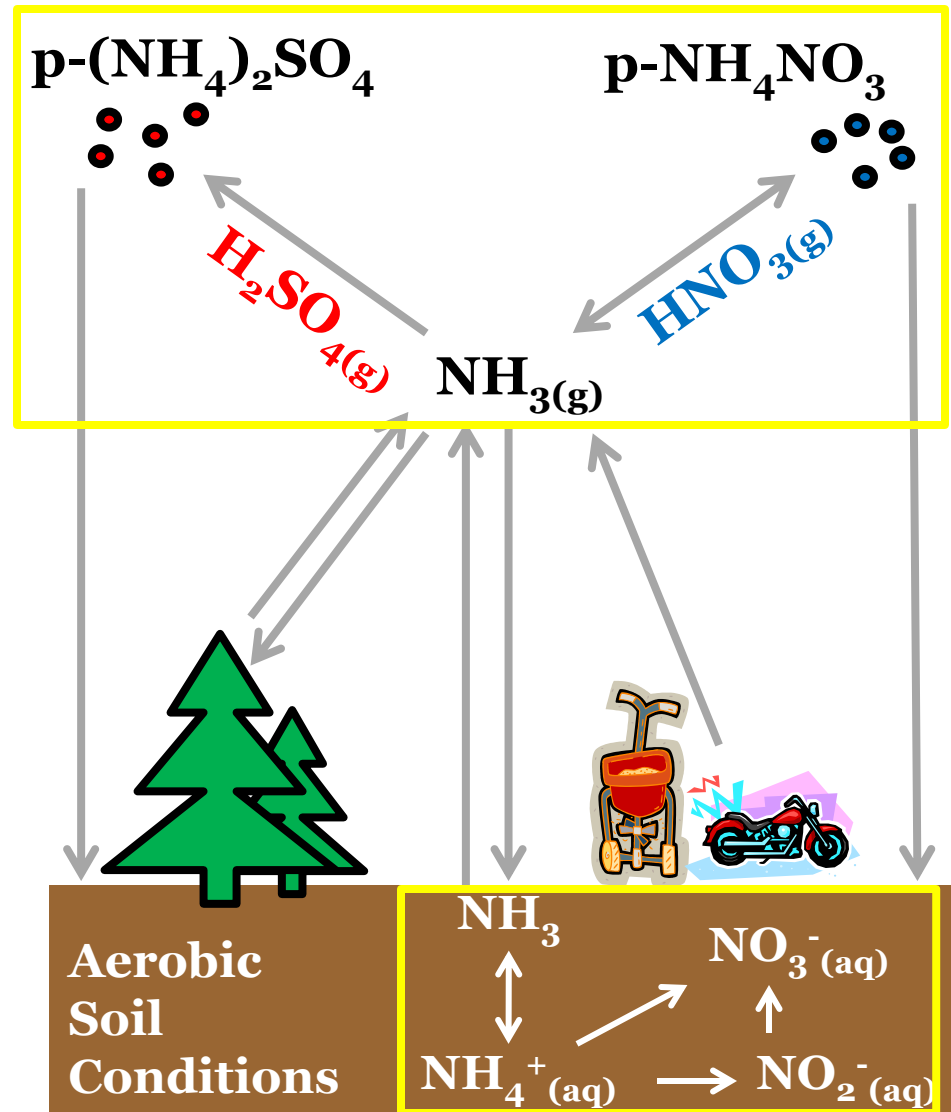
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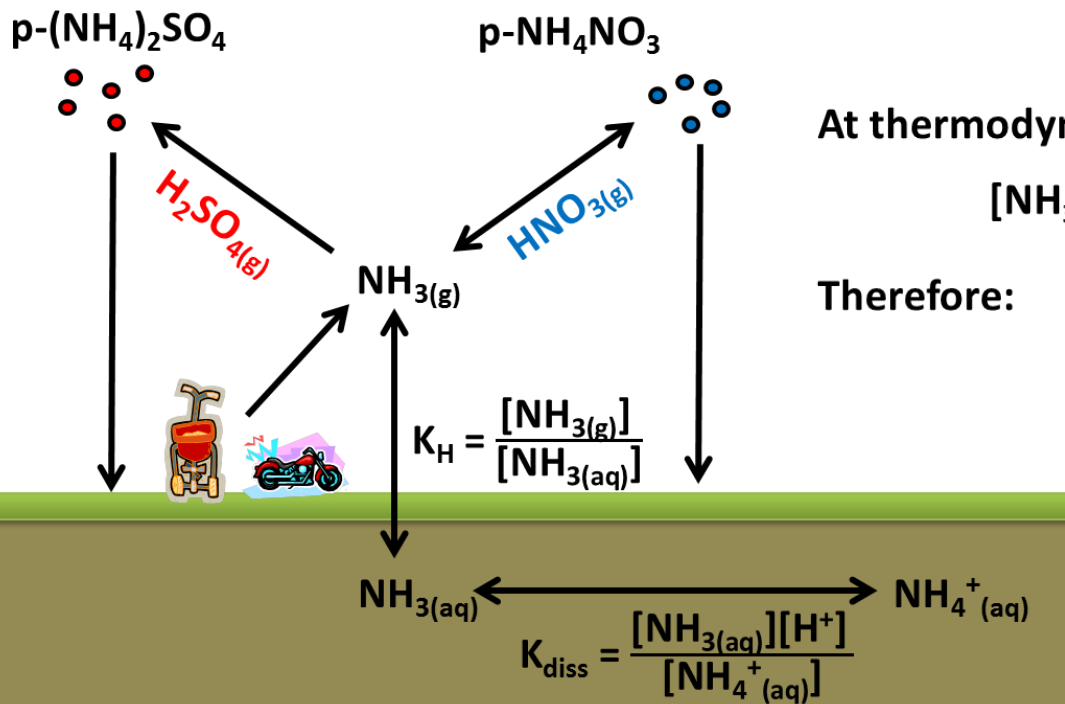
Ammonia in the Environment

- Atmosphere
 - Dominant alkaline gas
 - 2° aerosol formation
- Emissions
 - Mostly agricultural
- Deposition
 - Traditionally a “one-way street”
- Bidirectional Flux
 - “Two-way street”
 - Governed by compensation point (χ)
 - Poorly constrained
 - Lacking in most models



NH₃ Soil Compensation Point (χ_{soil})

- Equilibrium between NH_{3(g, atmo)} and NH_{4⁺(aq, soil)}
 - Dependent on soil temperature, pH and [NH_{4⁺]}
- If NH_{3(g)} < χ_{soil} , then soil emission
- Bash *et al.* included bi-directional flux in CMAQ
 - 10% increase in NH₃, 45% decrease in NH₃ dry deposition



At thermodynamic equilibrium:

$$[\text{NH}_3(\text{g})] = \frac{K_{\text{diss}}[\text{NH}_4^+(\text{aq})]}{K_H[\text{H}^+]}$$

Therefore:

$$\chi = \frac{K_{\text{diss}}[\text{NH}_4^+(\text{aq})]}{K_H[\text{H}^+]} e^{\left(\frac{\Delta H_{\text{vap}} + \Delta H_{\text{dis}}}{R}\right) \left(\frac{1}{298.15\text{K}} - \frac{1}{T_{\text{soil}}}\right)}$$

$$\chi = 13,587 \frac{[\text{NH}_4^+(\text{aq})]}{[\text{H}^+]} e^{\frac{-10,396\text{K}}{T_{\text{soil}}}} \times 10^9 \text{ ppb}$$

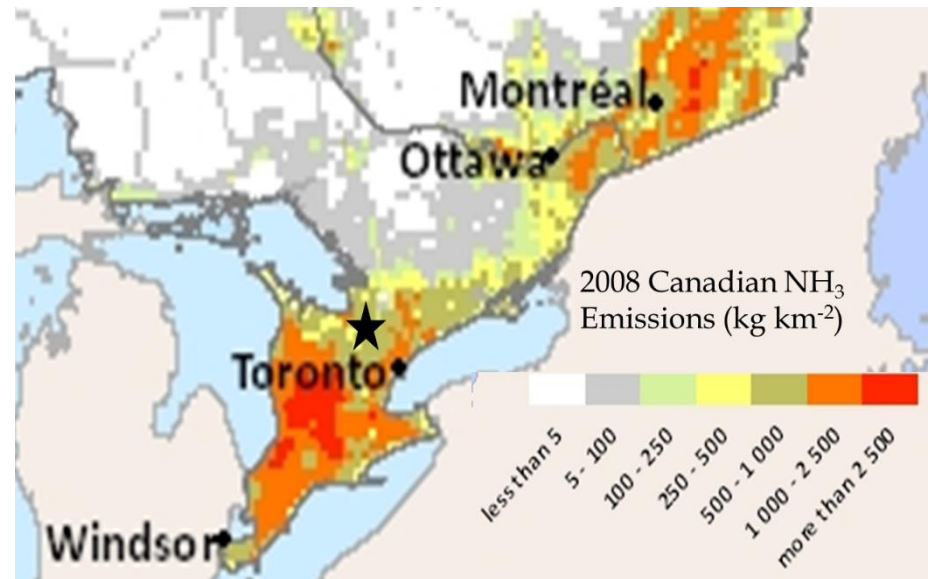
Bash et al. (2013) *Biogeosciences*, **10**, 1625.

Nemitz et al. (2004) *Atmos. Chem. Phys.*, **4**, 989.

CONTACT 2012

Characterizing Ontario Nitrogen Transport And Chemical Transformation

- Primary Objective
 - Provide observational constraints on χ_{soil} and NH_3 fluxes
- Motivation
 - Lack of studies over *non*-fertilized grasslands[‡]
 - Minimal field measurements of χ_{soil}
- Egbert, ON (★)
 - August 13 to October 2



[‡] Zhang et al. (2010) *J. Geophys. Rev.*, **115**, D20310.

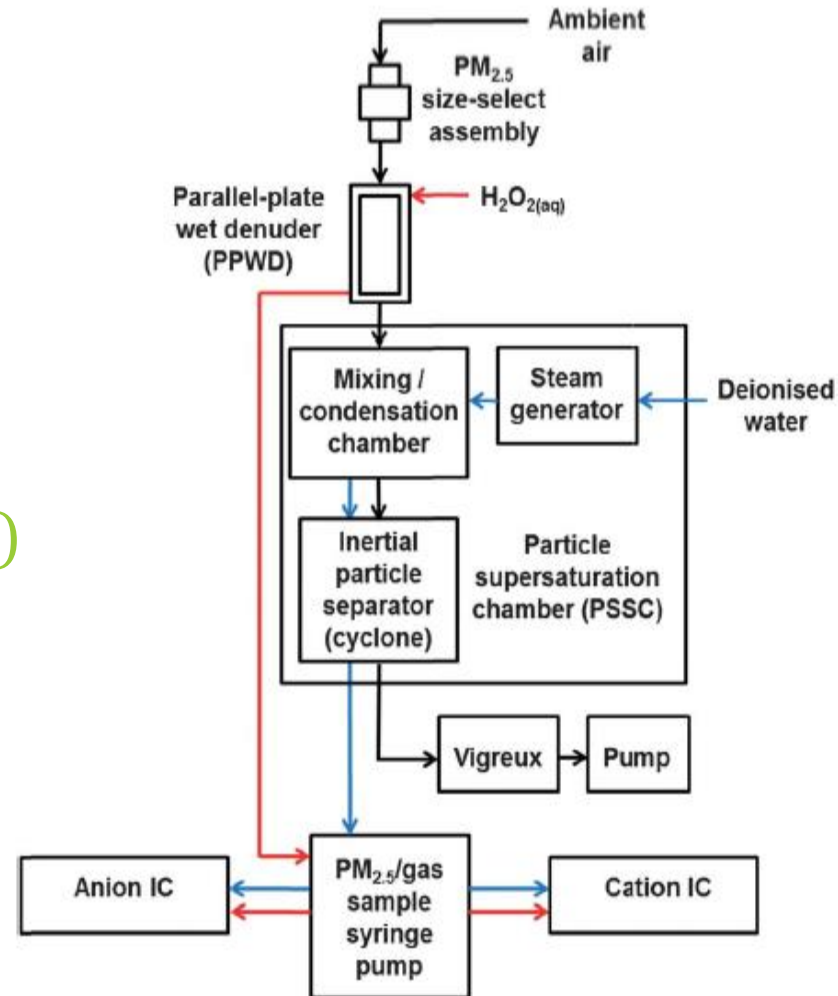
Methodology - Atmospheric Sampling

- Ambient Ion Monitor-Ion Chromatograph (AIM-IC)

- Simultaneous on-line quantification (via Ion Chromatography) of water soluble:

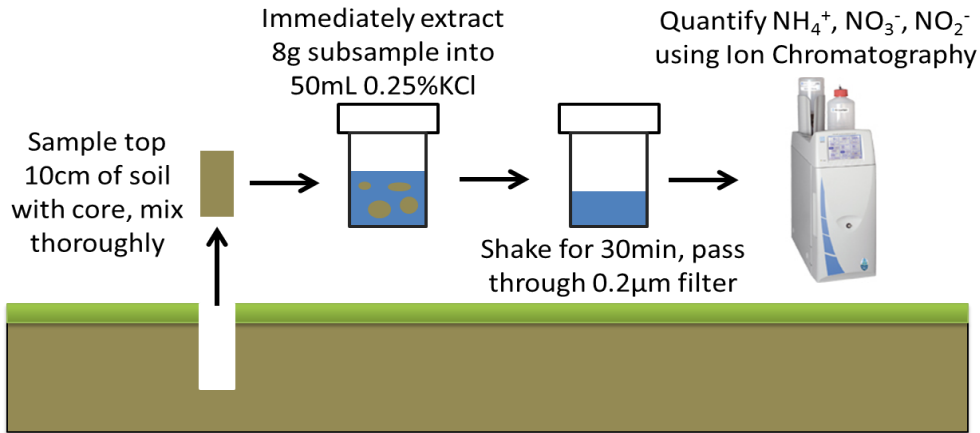
- Gas-phase species (NH_3 , HNO_3 , SO_2 , HCl , HONO , etc...)
- Ions in $\text{PM}_{2.5}$ (NH_4^+ , SO_4^{2-} , NO_3^- , Cl^- , NO_2^- , etc...)

- Hourly time resolution



Methodology - Soil Sampling

- Nitrogen Speciation



- Temperature/Water Content

- Hourly averages with commercial sensors

- pH

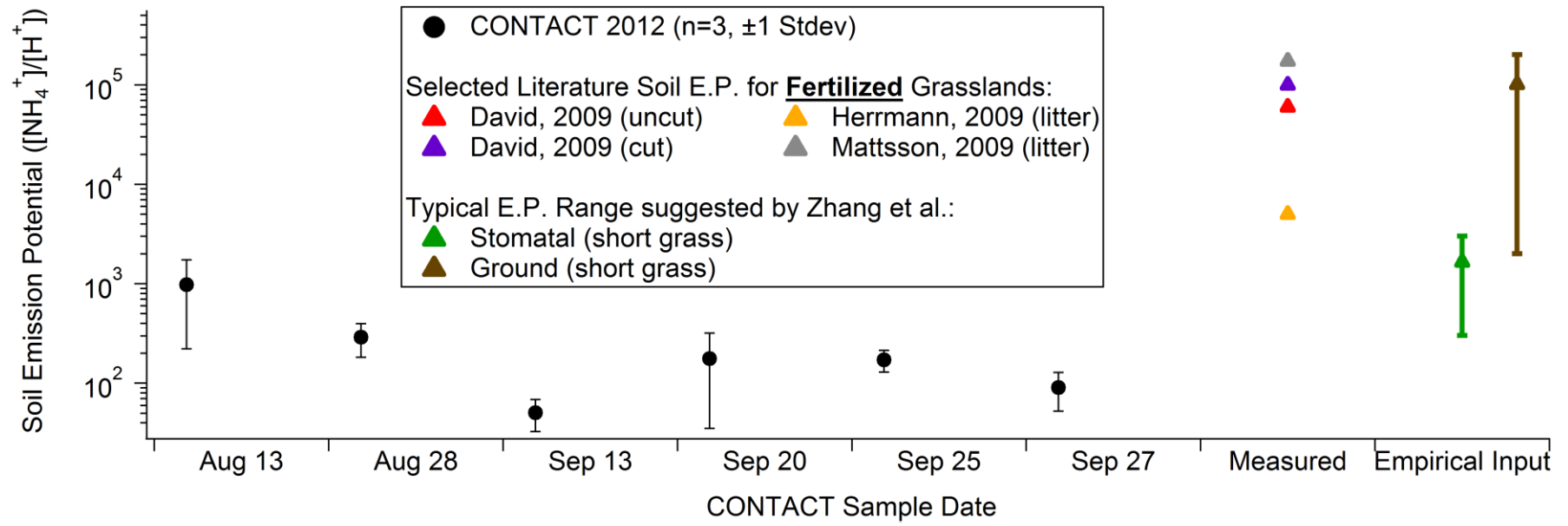
- pH electrode in 1:1 slurry of soil and deionized water

- Vegetation not sampled

- Small N-pool relative to soil

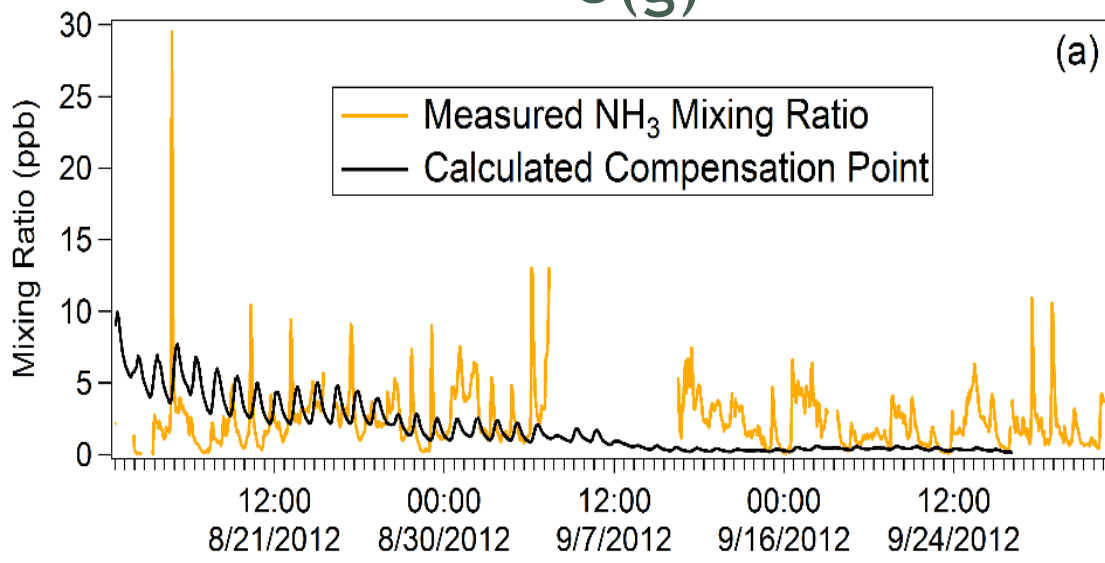
Results - Soil Emission Potential (Γ_{soil})

$$\Gamma_{soil} = \frac{[NH_4^+]}{[H^+]} \chi = 13,587 \frac{[NH_4^+_{(aq)}]}{[H^+]} e^{\frac{-10,396K}{T_{soil}}} \times 10^9 \text{ ppb}$$

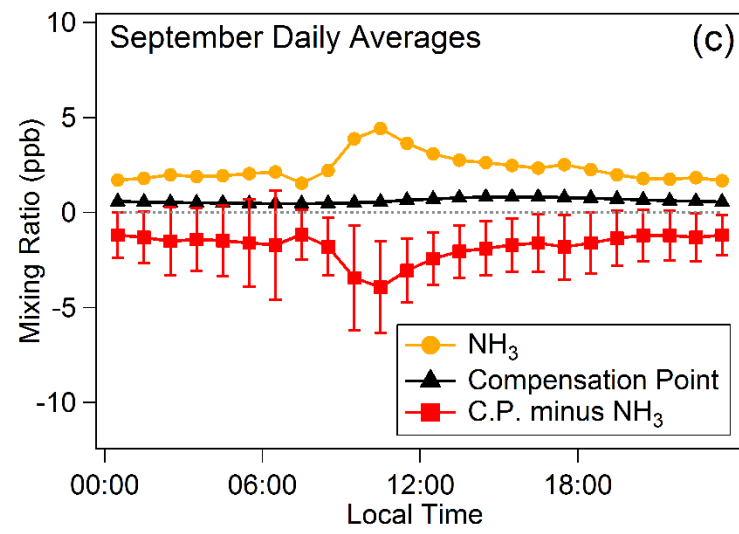
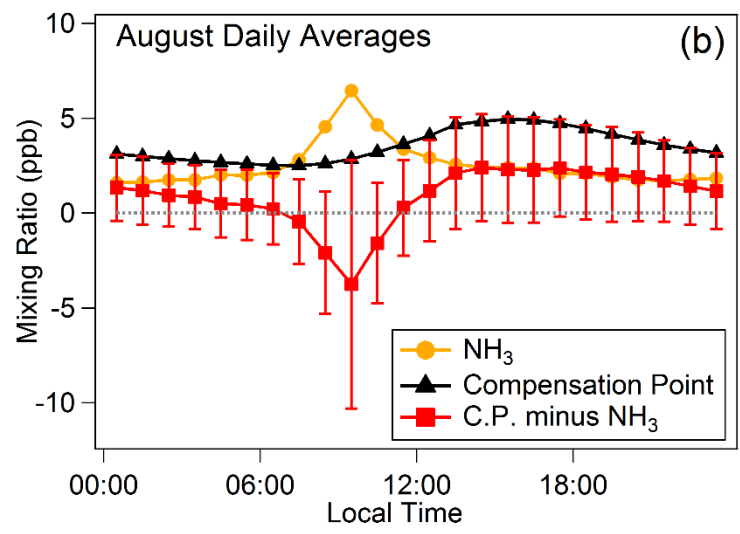


- First *measured* Γ_{soil} in a *non*-fertilized grassland
- Lower end of literature value for grasslands

Results - $\text{NH}_3(\text{g})$ and Calculated χ_{soil}



- Temporal trends in χ :
 - Seasonal (lower values in autumn)
 - Daily (lower values at dawn)



Results - Estimated Fluxes

- Assumptions:
 - z_0 is 0.05m
 - Soil resistance and vegetation are negligible
- Previously *measured* fluxes[†]:
 - 4 ng m⁻²s⁻¹ in summer
 - 24 ng m⁻²s⁻¹ in autumn

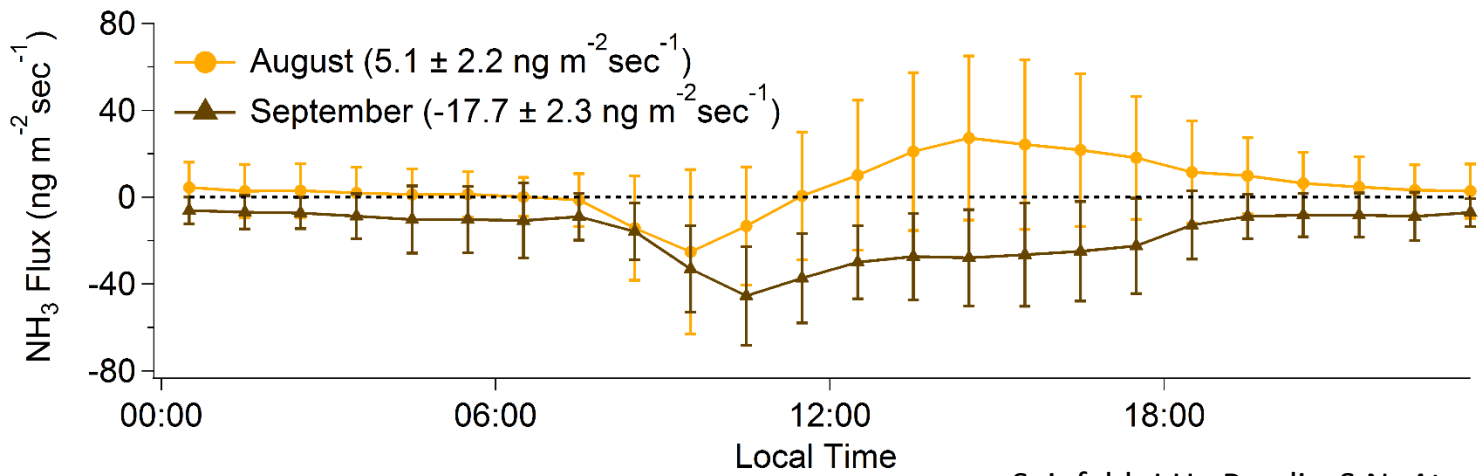
$$\text{Flux} = V_{\text{exchange}} * (\chi - \text{NH}_{3(g)})$$

$$V_{\text{exchange}} \approx \frac{1}{R_a + R_b}$$

$$R_a = \frac{\ln(z_{\text{ref}}) - \ln(z_0)}{\kappa u^*}$$

$$R_b = \frac{2}{\kappa u^*} \left(\frac{Sc}{Pr} \right)^{\frac{2}{3}}$$

V_{exchange} = exchange velocity
 R_a = aerodynamic resistance
 R_b = quasi-laminar resistance
 z_{ref} = AIM-IC inlet height
 z_0 = 0.05m for uncut grass
 κ = 0.4
 u^* = friction velocity
 Sc = 1.07
 Pr = 0.72

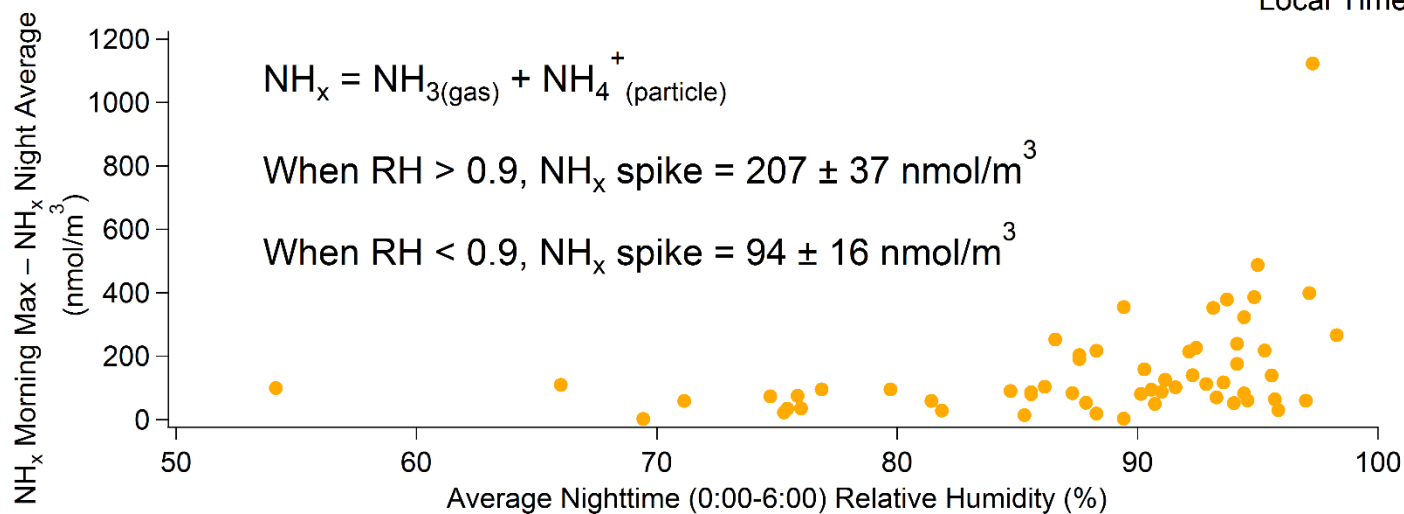
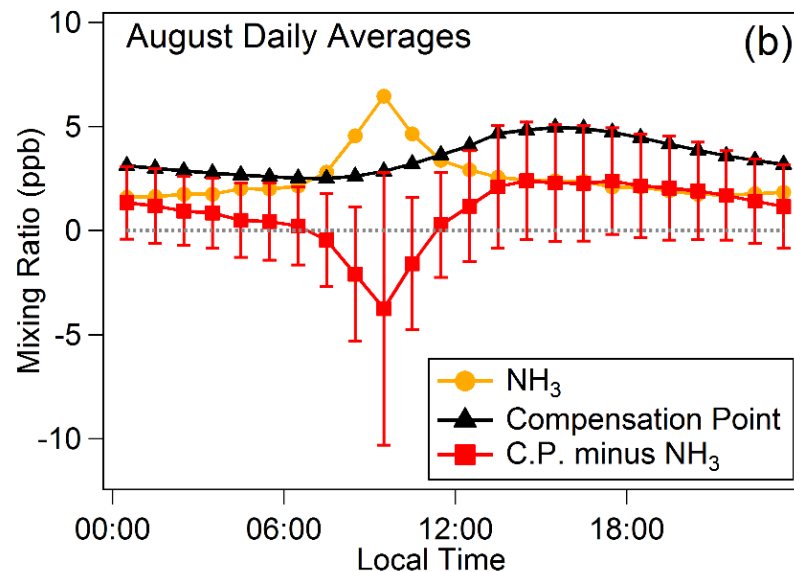


Seinfeld, J.H., Pandis, S.N. *Atmospheric Chemistry and Physics*. (Wiley, New York, 2006).

[†]Kruit et al. (2007) *Atmos. Environ.*, **41**, 1275.

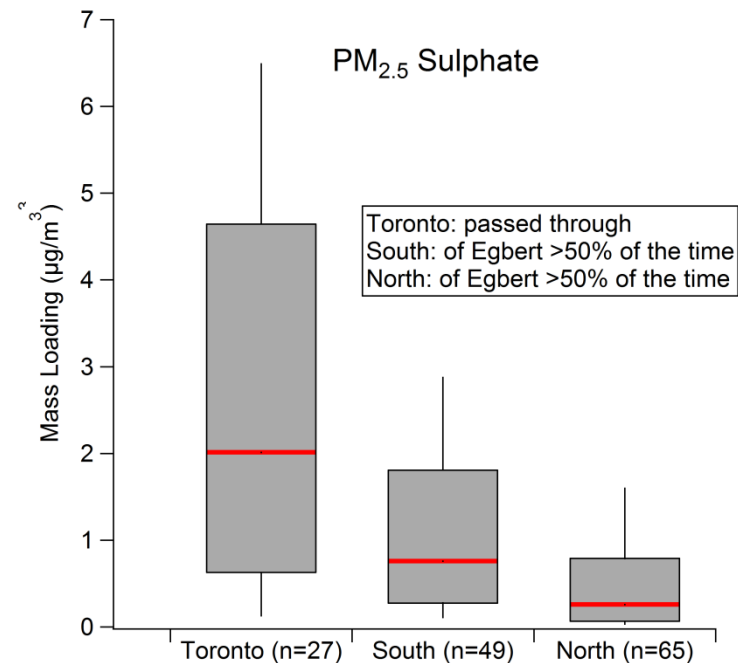
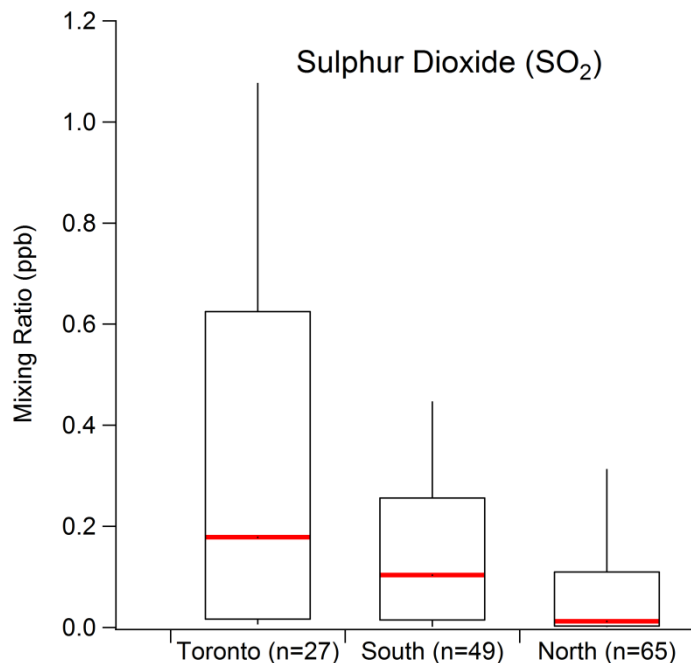
Results - NH₃ Morning Spike

- Previous studies (i.e. Ellis *et al.*) observe morning peak of NH₃ at ~10:00
- Possibly due to:
 - Volatilization of dew/fog
 - Plant/soil emissions
 - Nocturnal BL break up



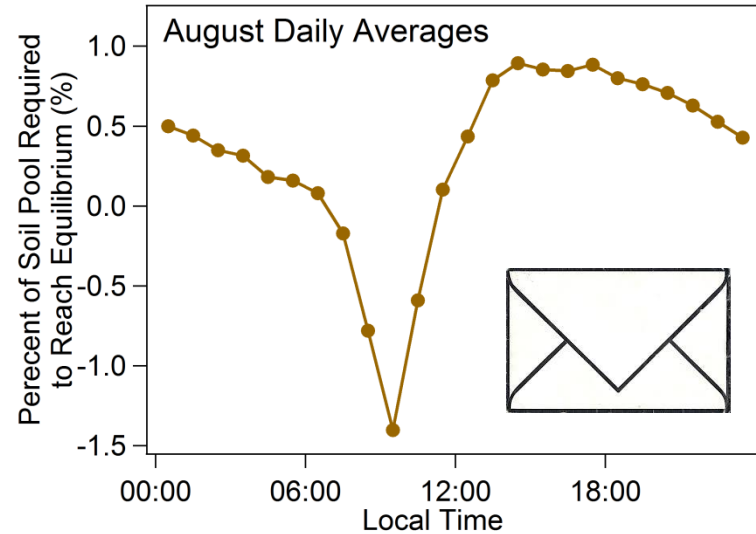
Bi-directional Exchange in Action?

- Used HYSPLIT to determine 48 hour back-trajectories
 - Calculated every 6 hours during CONTACT
 - Clear directional bias for PM_{2.5} and precursor gases
 - *Except* NH₃
 - Spatial homogeneity of local sources or evidence for bi-directional exchange?



Bi-directional Exchange in Action?

- Is the soil pool big enough?
 - Average fluxes during August
 - Boundary layer is 1000m
 - Top 10cm of soil exchanges
 - Soil is 1.5 g/cm³
- Is the exchange fast enough (how long to reach 50% equilibrium)?
 - Using v_{ex} and $(\chi_{soil} - NH_3)$ from August



Height (m)	14:00 (fast)	01:00 (slow)
1000	~11 hours	~37 hours
3	~2 min	~7 min

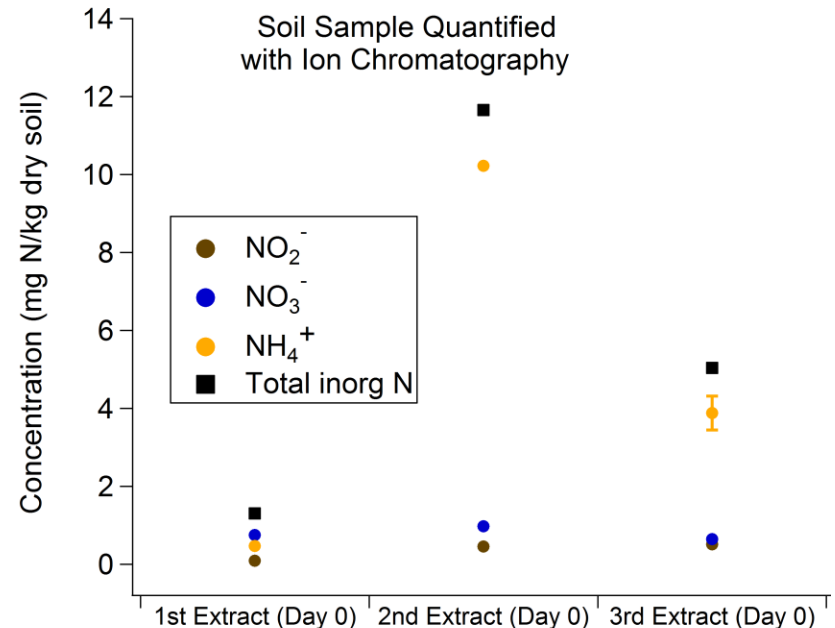
Difficulties in Studying NH_3 Fluxes

1. Measure Fluxes

- Typically requires intensive field work (unrealistic over large area)

2. Estimate Fluxes from surface and atmospheric properties (i.e. T, $[\text{NH}_4^+]$, pH, μ , NH_3 , etc...)

- Logistically easier
- Complicated by heterogeneity
- What methods are most representative of available NH_4^+ and H^+ ?



Conclusions

1. Γ_{soil} (non-fertilized grassland) is well below previous measurements for fertilized grasslands
2. Estimated Soil Fluxes:
 - Exhibit clear diurnal and seasonal trend
 - 5 and -18 ng m⁻²s⁻¹ for Aug and Sept, respectively
 - Consistent with previous measurements from Kruit *et al.*
3. Evidence that emissions from non-fertilized grasslands could sufficiently modulate *near-surface* NH₃ despite low Γ_{soil}
4. Morning Spike of NH₃ is likely related to dew/fog

Acknowledgements

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