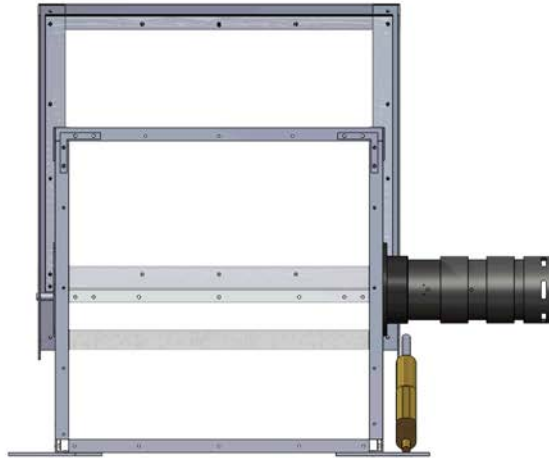


Development of the Next Generation of Flux-Measurement Tools



Berkeley B. Almand, Michael P. Hannigan, Nicolas B. Masson, Ricardo A. Piedrahita, Gregory Miller, Allison Moore, Kevin Klinkel, Alex Demarias

Mechanical Engineering Department, University of Colorado

John Ortega

Atmospheric Chemistry Division, National Center for Atmospheric Research

Eladio M. Knipping

Electric Power Research Institute



College of Engineering
& Applied Science



Dry Deposition is an important contributor to acidification

- The EPA is considering changing future secondary National Ambient Air Quality Standards (NAAQS) for SO₂ and NO₂ based on acidification
- SO₂ dry deposition accounts for up to 50% of total sulfur deposition
- Dry deposition accounts for up to 40% of total nitrogen deposition
- There is a lot of uncertainty regarding dry deposition measurements and modeling

Flux-measurement methods are costly or indirect



Inferential Method:

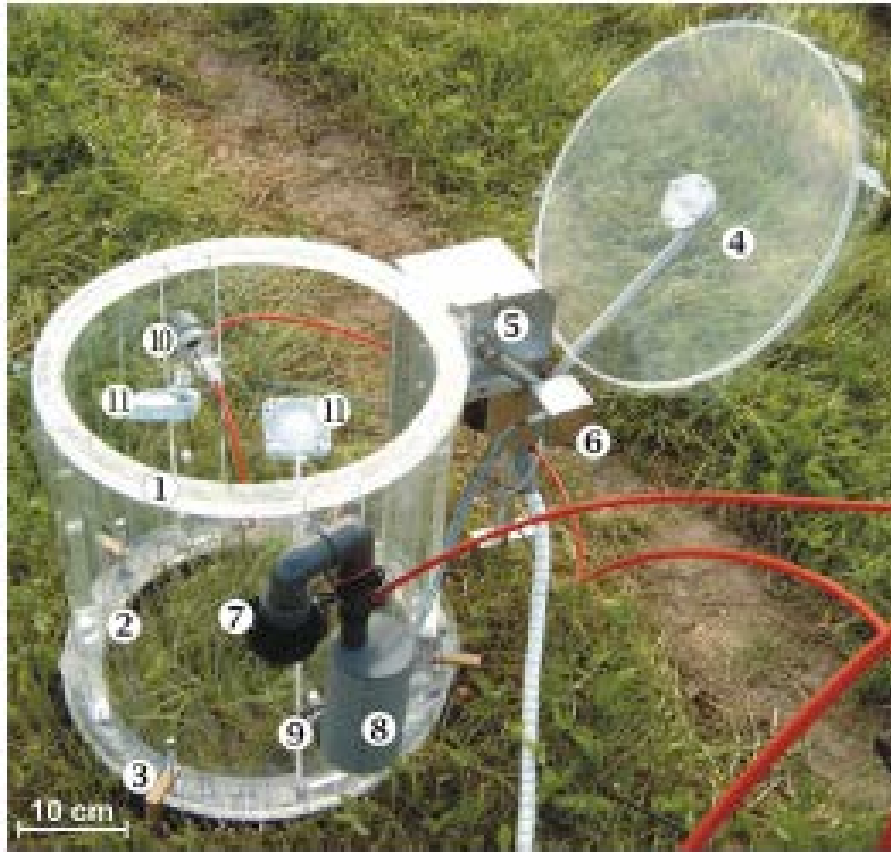
Advantages:

- Less expensive than eddy direct measurement
- covariance
- Most accurate method available
- Less technically difficult

Disadvantages:

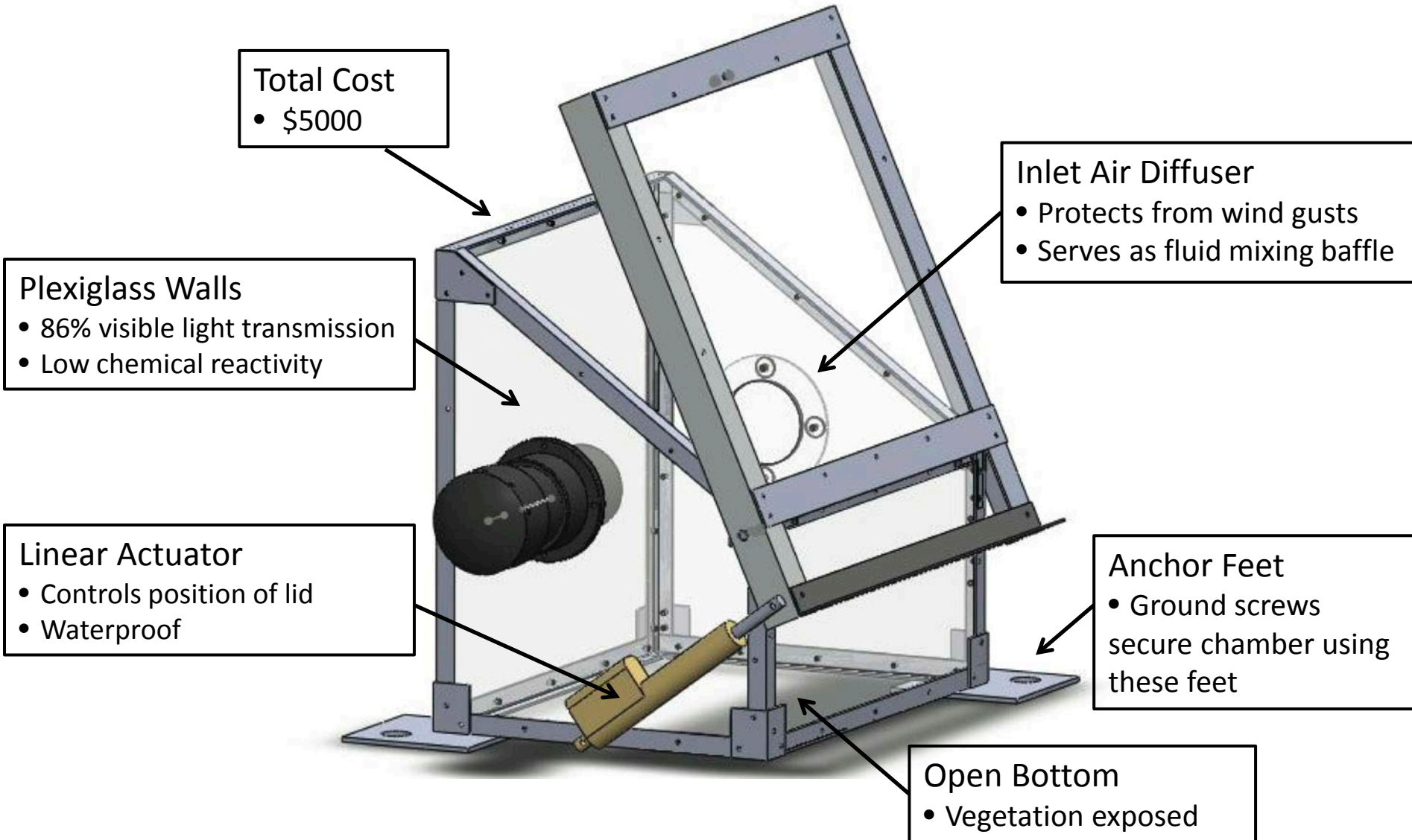
- Based on models that don't always agree with direct measurements, or other models
- Expensive (> \$500,000)
- Technically difficult and computationally expensive
- Flux estimates between models can vary by a factor of 2 to 3

Pape et al. demonstrated that flux chambers are an accurate tool for measuring dry deposition



- Dynamic flux chambers were used in combination with traditional pollutant monitors
- Measured CO₂ and methanol surface flux over grassland
- Demonstrated good agreement with eddy covariance systems

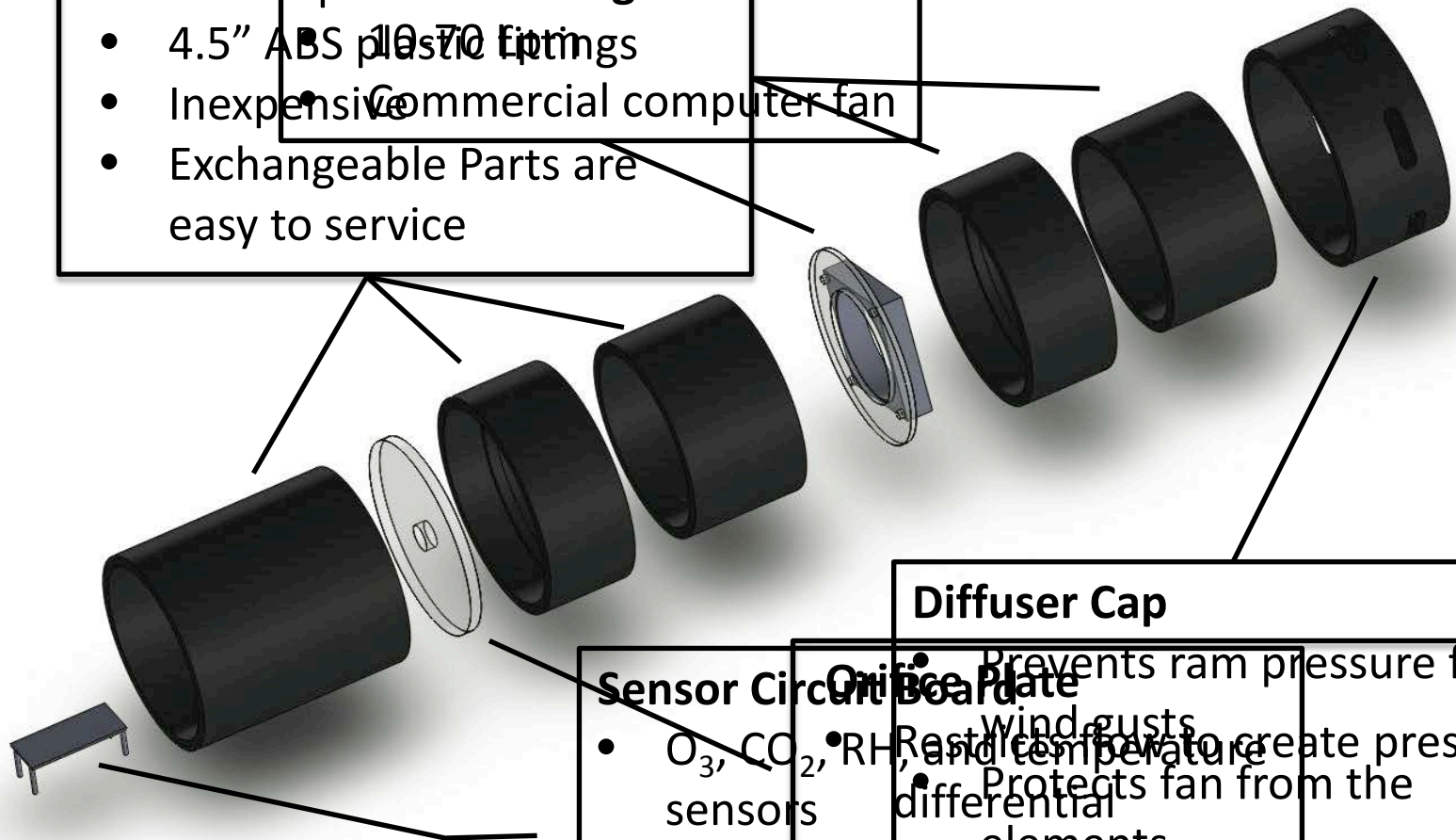
Dynamic Flux Chambers could provide direct dry-deposition measurements



Chamber outlet houses sensors and flow controls

Standard Components

- 4" ABS Fan & Mounting Plate
- 4.5" ABS plastic fittings
- Inexpensive Commercial computer fan
- Exchangeable Parts are easy to service



Diffuser Cap

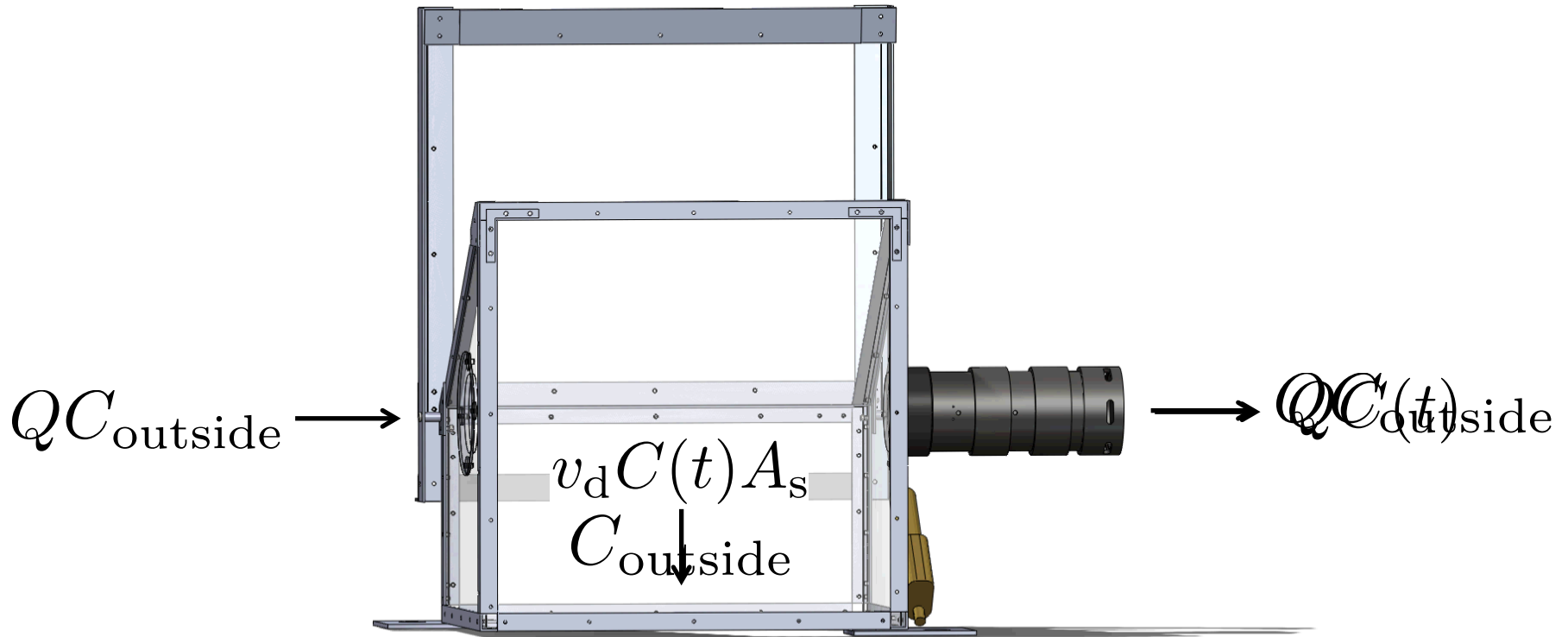
- Prevents ram pressure from wind gusts
- Rests on top to create pressure differential
- Protects fan from the elements
- Enables pressure transducer to measure flow rate

Sensor Circuit Board

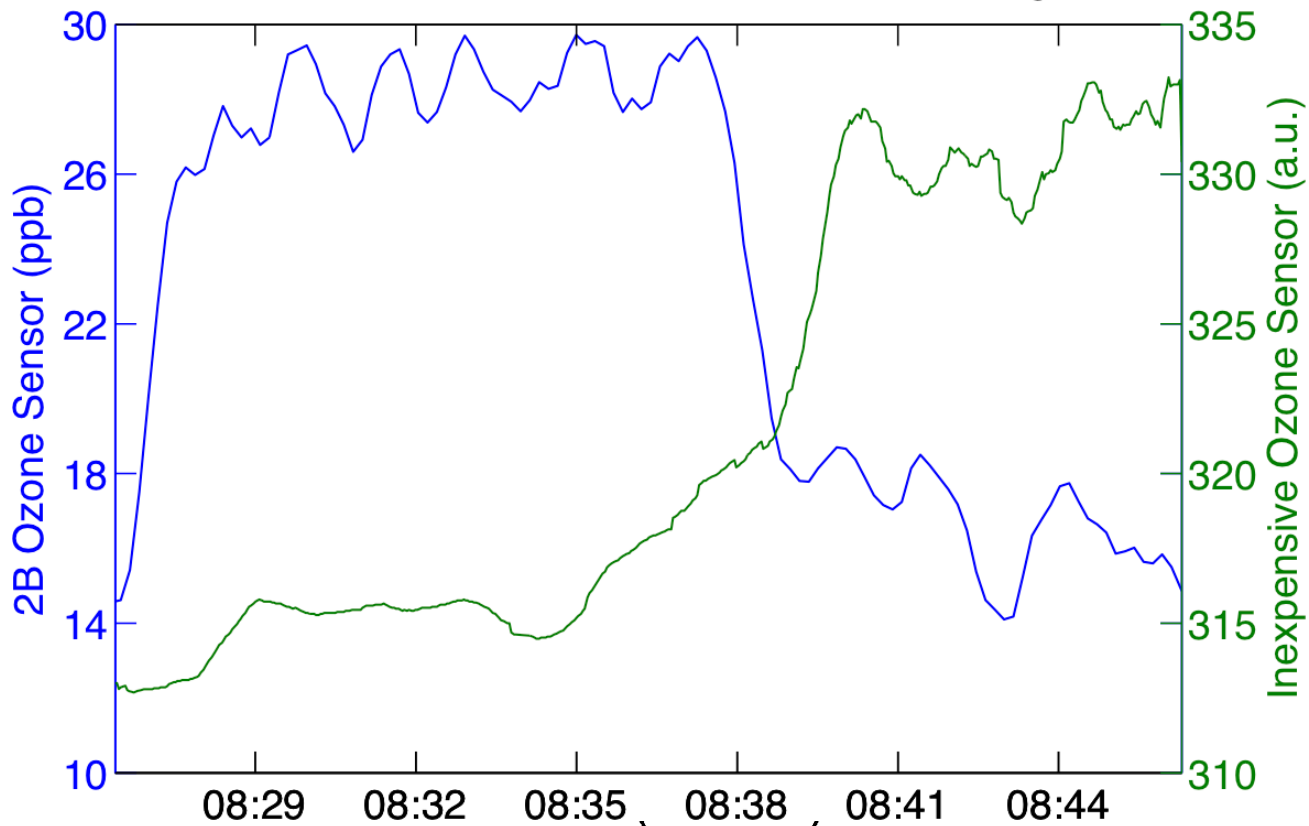
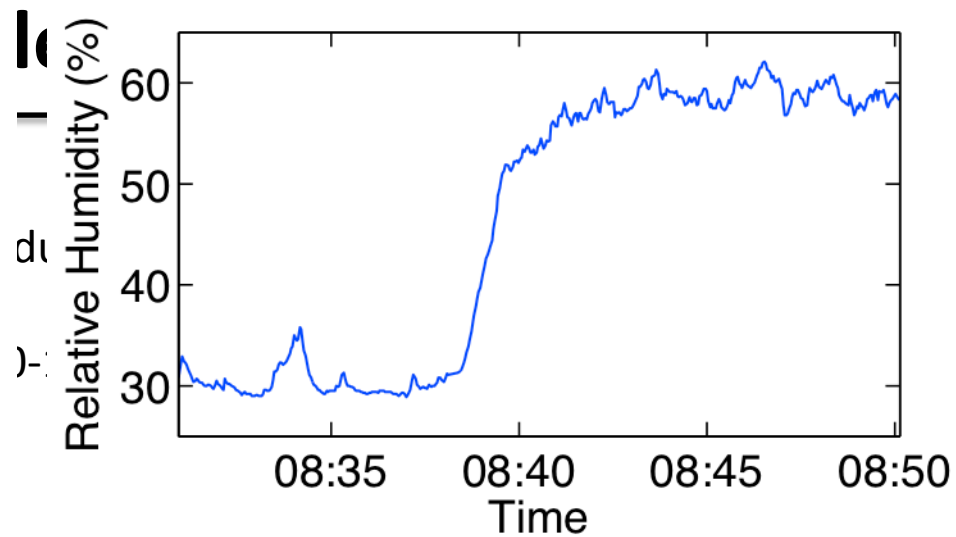
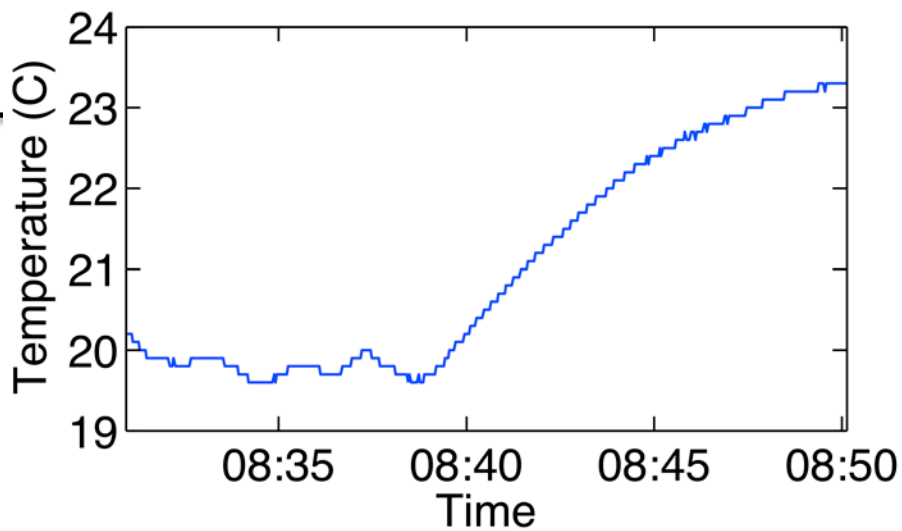
- O₃, CO₂, RH, and temperature sensors
- mounted on same board
- place sensors in center of flow path



Dry deposition flux is calculated via a mass balance



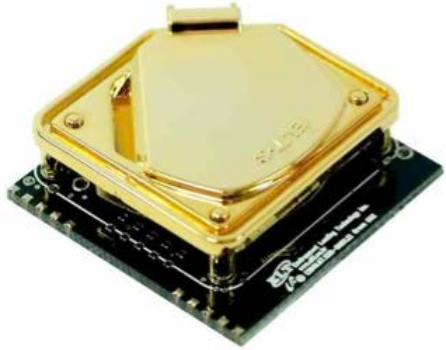
$$C(t) = \frac{QC_{\text{out}}}{Q - v_d A_s} + \left(C_{\text{out}} - \frac{QC_{\text{out}}}{Q - v_d A_s} \right) e^{-\frac{(Q - v_d A_s)t}{V}}$$



Plot Parameters

- $C_{\text{ozone}} = 30 \text{ ppb}$
- $v_d = -0.1 \text{ cm/s}$

Inexpensive sensors enable low-cost measurements

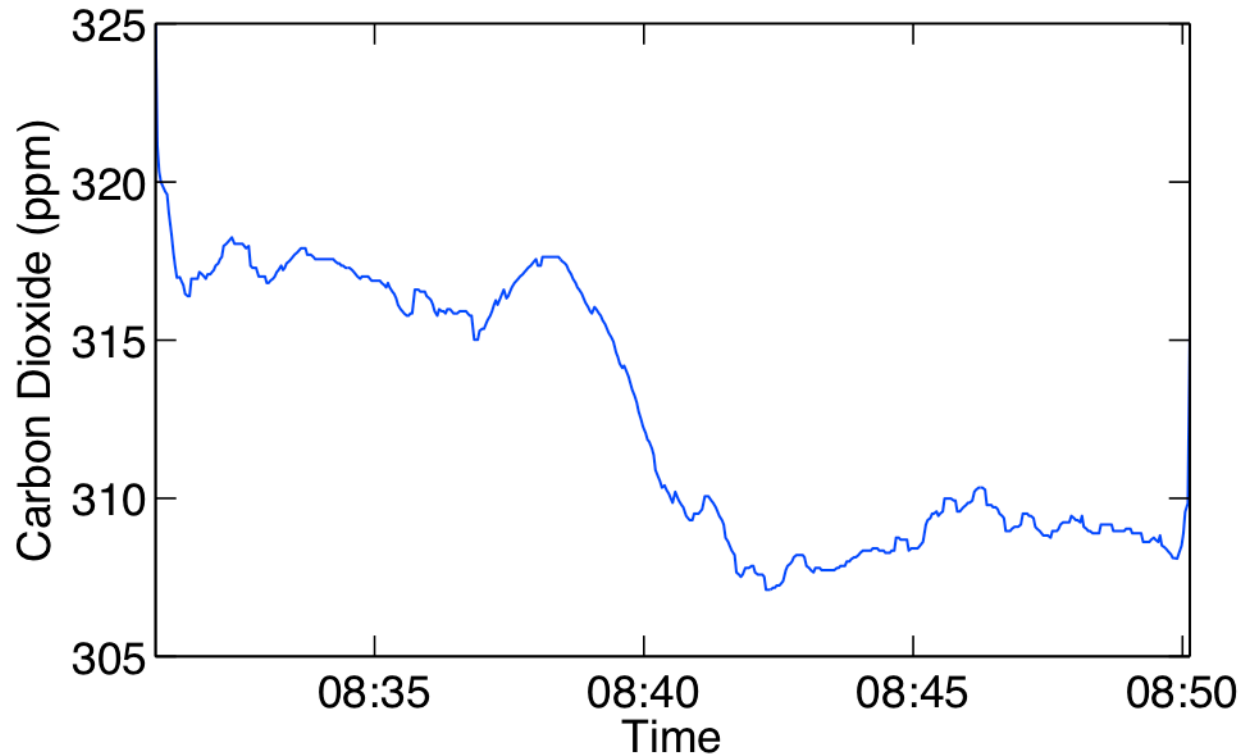


CO₂ Sensor

- Non-dispersive infrared
- Costs about \$60
- Detection range of 0-5000 ppm
- 30 ppm resolution

Concerns

- Sensitive to temperature



Inexpensive sensors enable low-cost measurements



NO₂ Sensor

- Electrochemical Sensor
- Costs about \$80
- Detection range of 0-20,000 ppb
- < 20 ppb resolution

Concerns

- Sensor resolution (ideally 1 ppb)
- NO emissions react with O₃ and rapidly produce NO₂. This can diminish the magnitude of NO₂ deposition reading



SO₂ Sensor

- Electrochemical Sensor
- Costs about \$80
- Detection range of 0-50,000 ppb
- < 100 ppb resolution

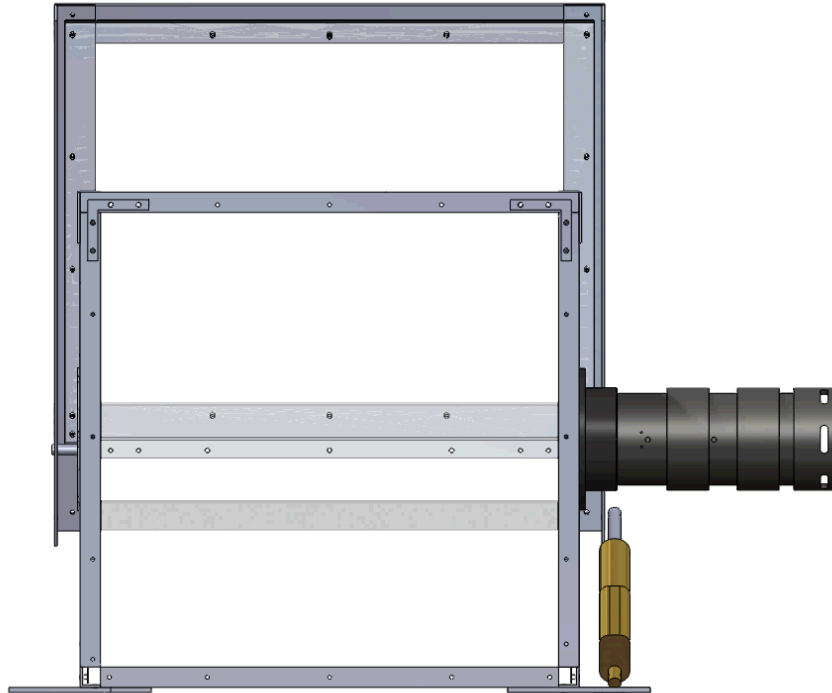
Concerns

- Sensor resolution (ideally 1 ppb)
- SO₂ levels in the western US are low, so SO₂ deposition will be very difficult to measure

Summary & Future Plans

- We developed a flux chamber that measures CO₂, O₃, RH, temperature, rainfall, and soil moisture.
- The crux of our project is finding inexpensive ways to take high-resolution NO₂ and SO₂ measurements.
- We will develop electronics and install the NO₂ and SO₂ sensors. We will also install high-resolution NO₂ measurement devices, which will enable us to calibrate and evaluate the inexpensive sensors.
- We will perform calibrations to explore the effects of temperature, RH, and cross-sensitivity on the O₃ sensors.
- We will compare our flux-chamber results to an eddy-covariance system.

Questions?

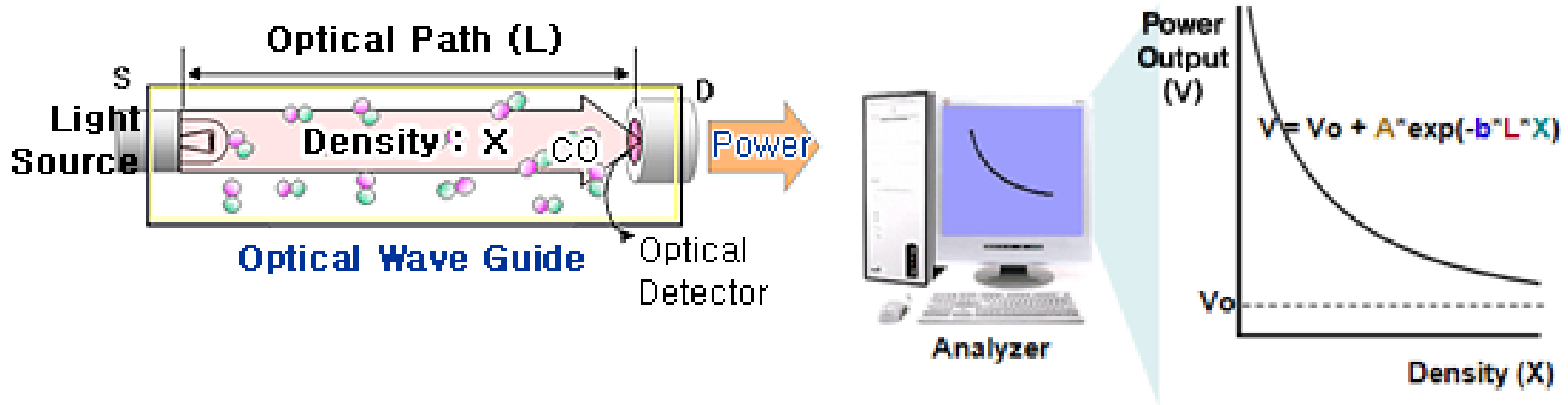


Contact Information: Berkeley Almand, Mechanical Engineering
Department, University of Colorado, berkeley.almand@colorado.edu

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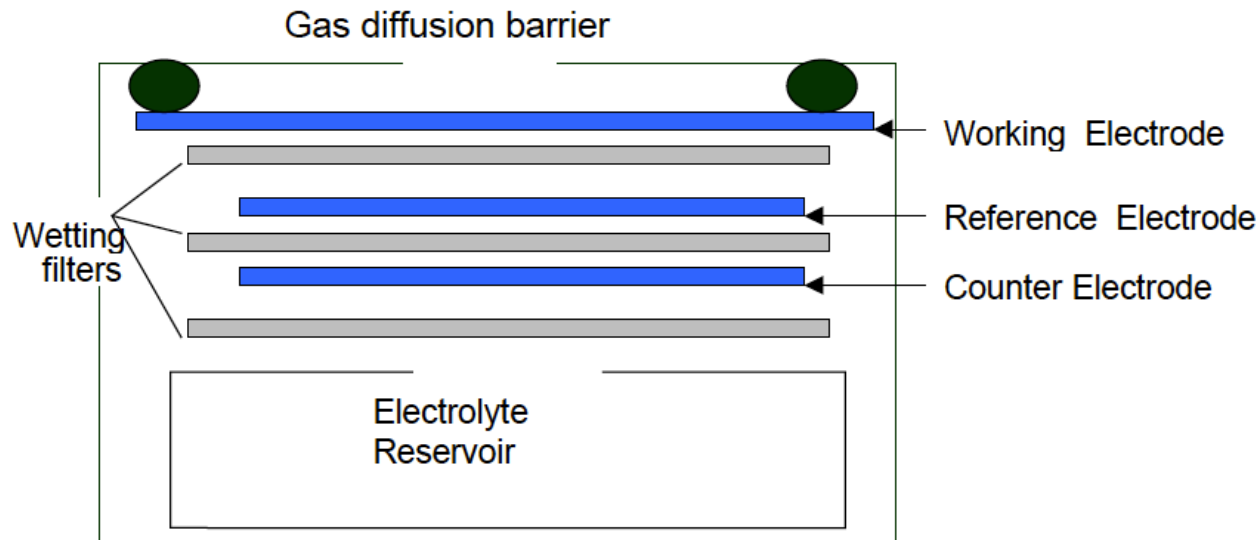
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Non-dispersive infrared radiation (NDIR)



- NDIR: Infrared light is directed through a sample chamber toward a detector.
- Each gas absorbs infrared radiation at a different wavelength (CO_2 absorbs $4.26\mu\text{m}$).
- Concentration (density) can be calculated from measured voltage and optical path length.

Electrochemical Sensors



- Measure concentration by oxidizing or reducing gas and measuring current.

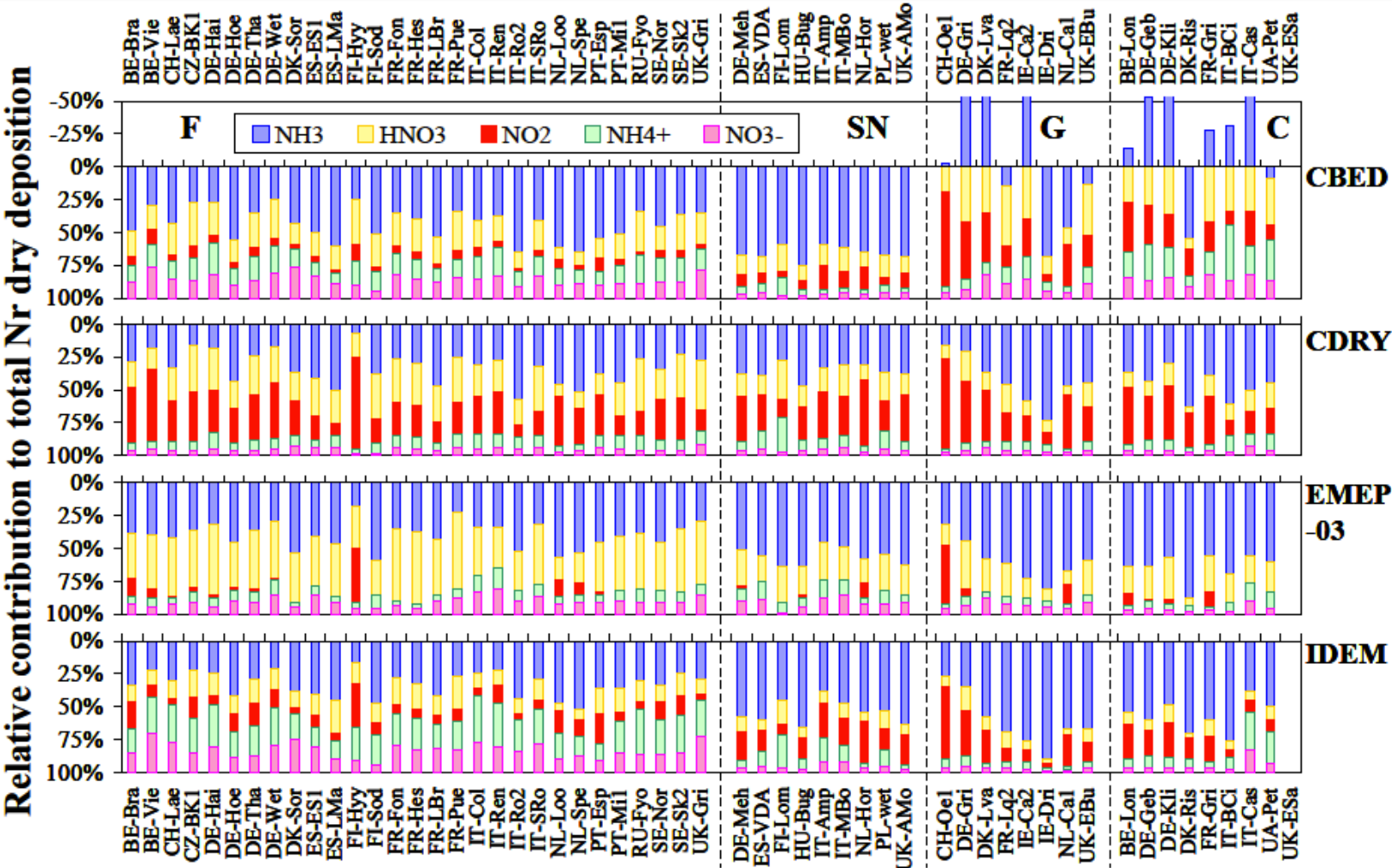
Advantages:

- Output is linearly proportional to concentration
- Stable over time (less re-calibration)

Disadvantages:

- Cross-sensitive

Relative contributions of N_r species to total inorganic N dry deposition



*Negative percentages for NH₃ denote net NH₃ emissions, which are expressed relative to the sum of dry deposition fluxes for the other four N_r species.

Chemiluminescence

Nitrogen Chemistry
