



**Trends in Sulfate
Deposition and
Stream Water
Base Cation
Concentrations in
the Catskill
Mountains of
New York**



Catskill Long-Term Monitoring



Funded by:





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Greg Lawrence
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New York City's Water Supply System

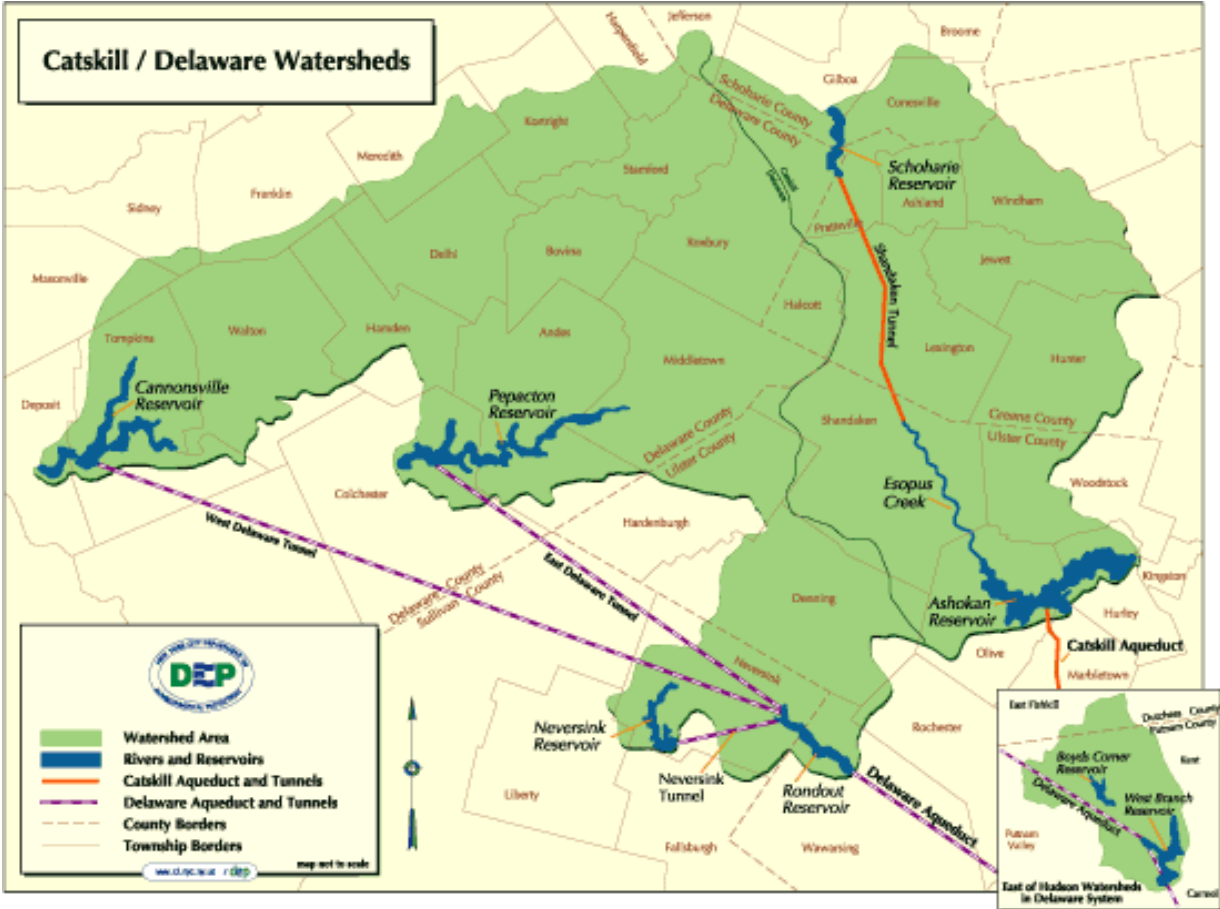


DEP
NEW YORK CITY DEPARTMENT OF ENVIRONMENTAL PROTECTION

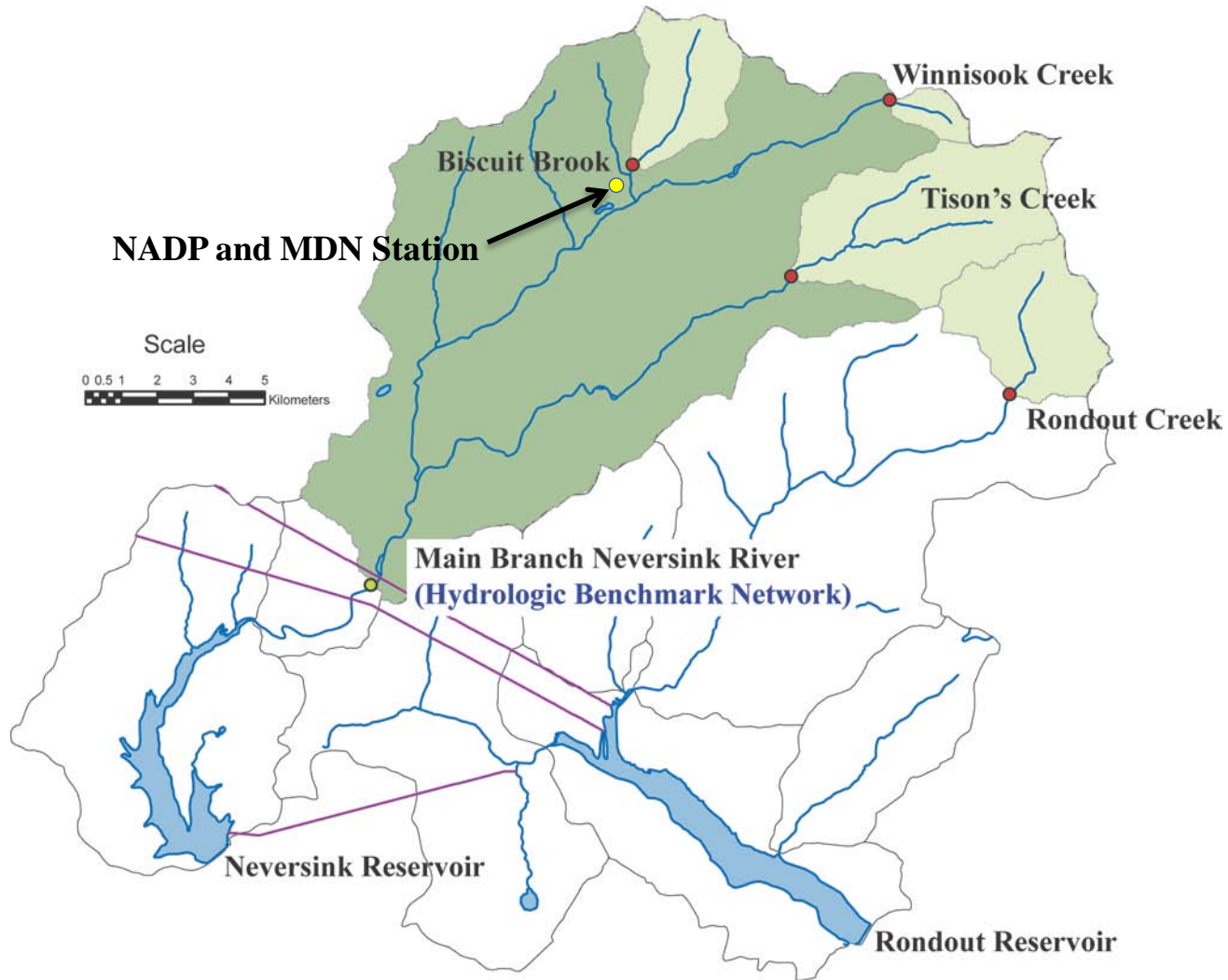
- Catskill / Delaware Watershed Area
- Croton Watershed Area
- Rivers and Reservoirs
- Catskill Aqueduct and Tunnels
- Croton Aqueduct
- Delaware Aqueduct and Tunnels
- County Borders
- State Borders

www.nyc.gov/dep

Catskill / Delaware Watersheds



Catskill LTM Network



Study Area

- ◆ **Geology:** uplifted plateau of sedimentary bedrock (sandstone, conglomerate and some shale and siltstone), covered by variably thick layer of till from the most recent glaciation.
- ◆ **Topography:** steep slopes, dissected by streams
- ◆ **Soils:** thin till and soil cover, soils are inceptisols: bouldery silt loam with thin organic horizons
- ◆ **Vegetation and land use:** 99% forested, dominated by American Beech, Sugar Maple, Yellow Birch

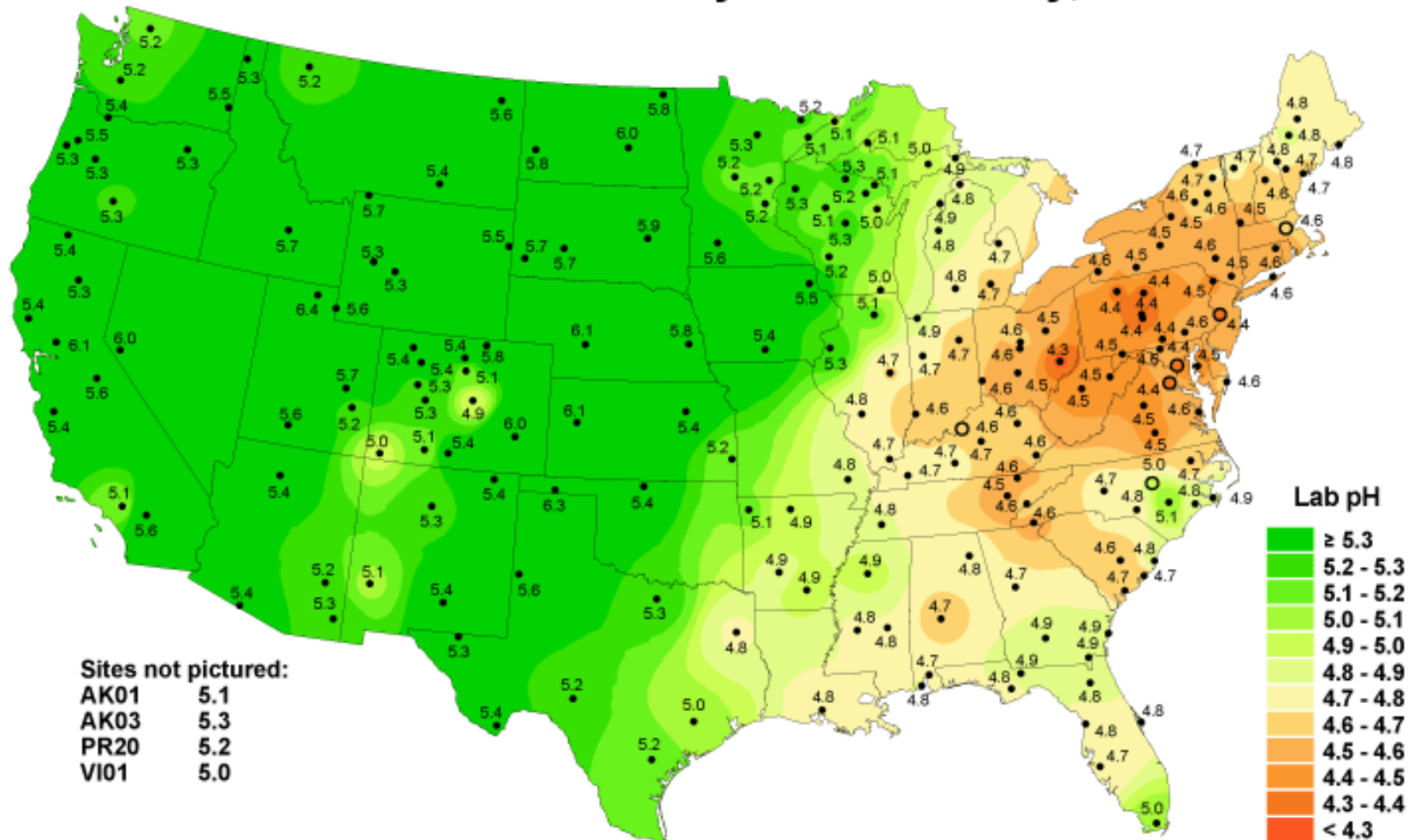


Winnisook Watershed (WN) on the slopes of Slide Mountain in the headwaters of the Neversink River basin



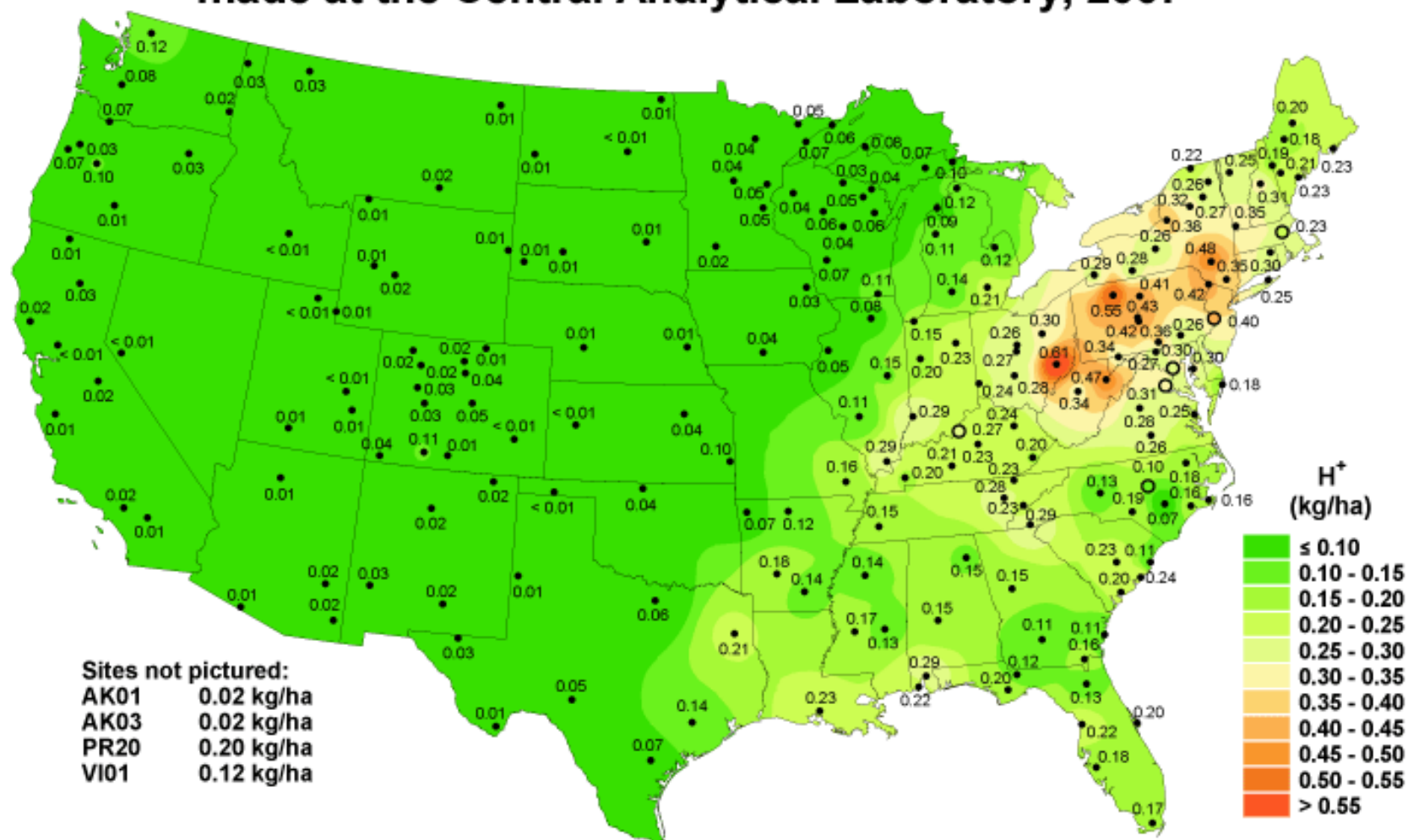
Rondout Creek above Peekamoose (RC)

Hydrogen ion concentration as pH from measurements made at the Central Analytical Laboratory, 2007



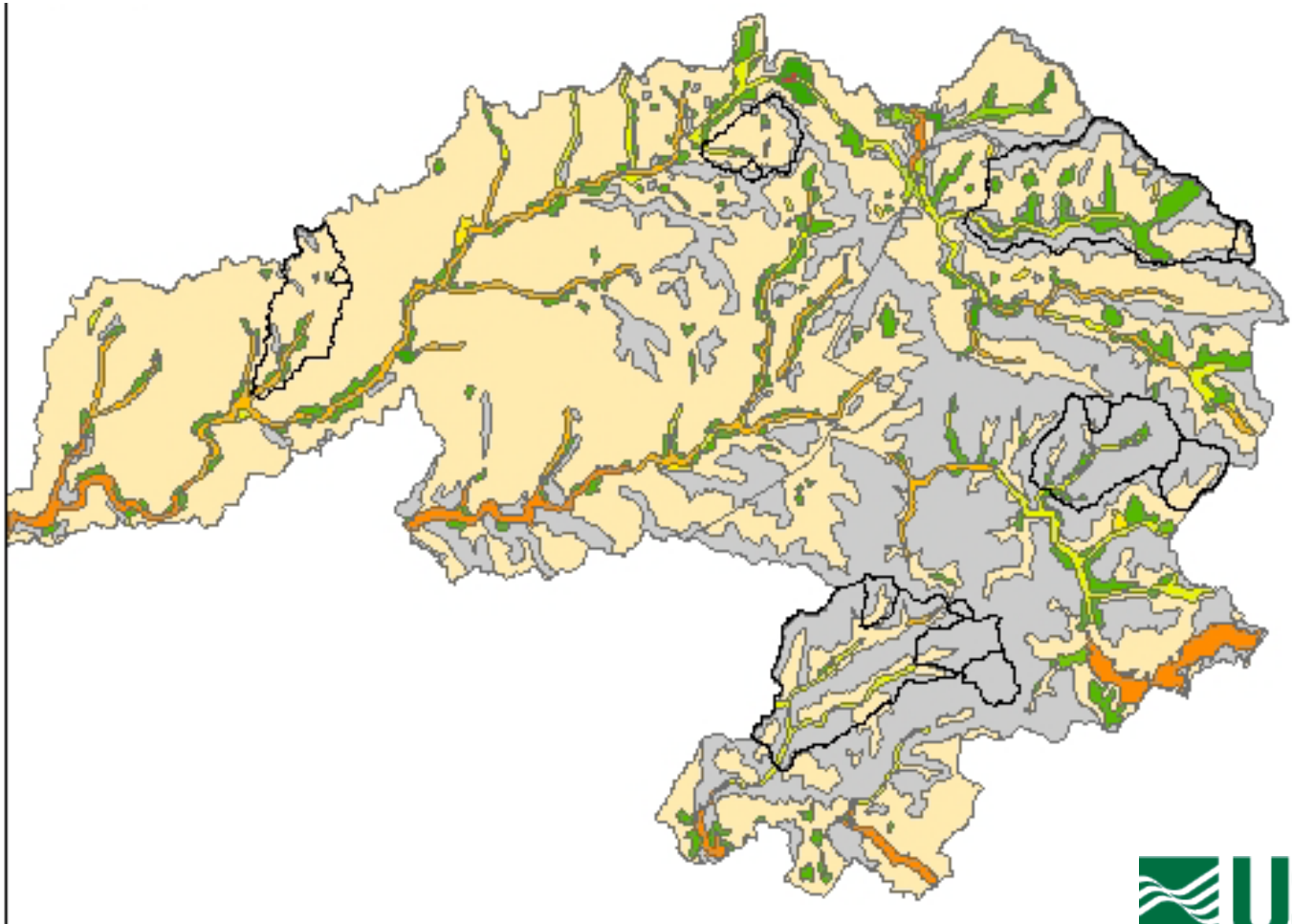
National Atmospheric Deposition Program/National Trends Network
<http://nadp.sws.uiuc.edu>

Hydrogen ion wet deposition from measurements made at the Central Analytical Laboratory, 2007



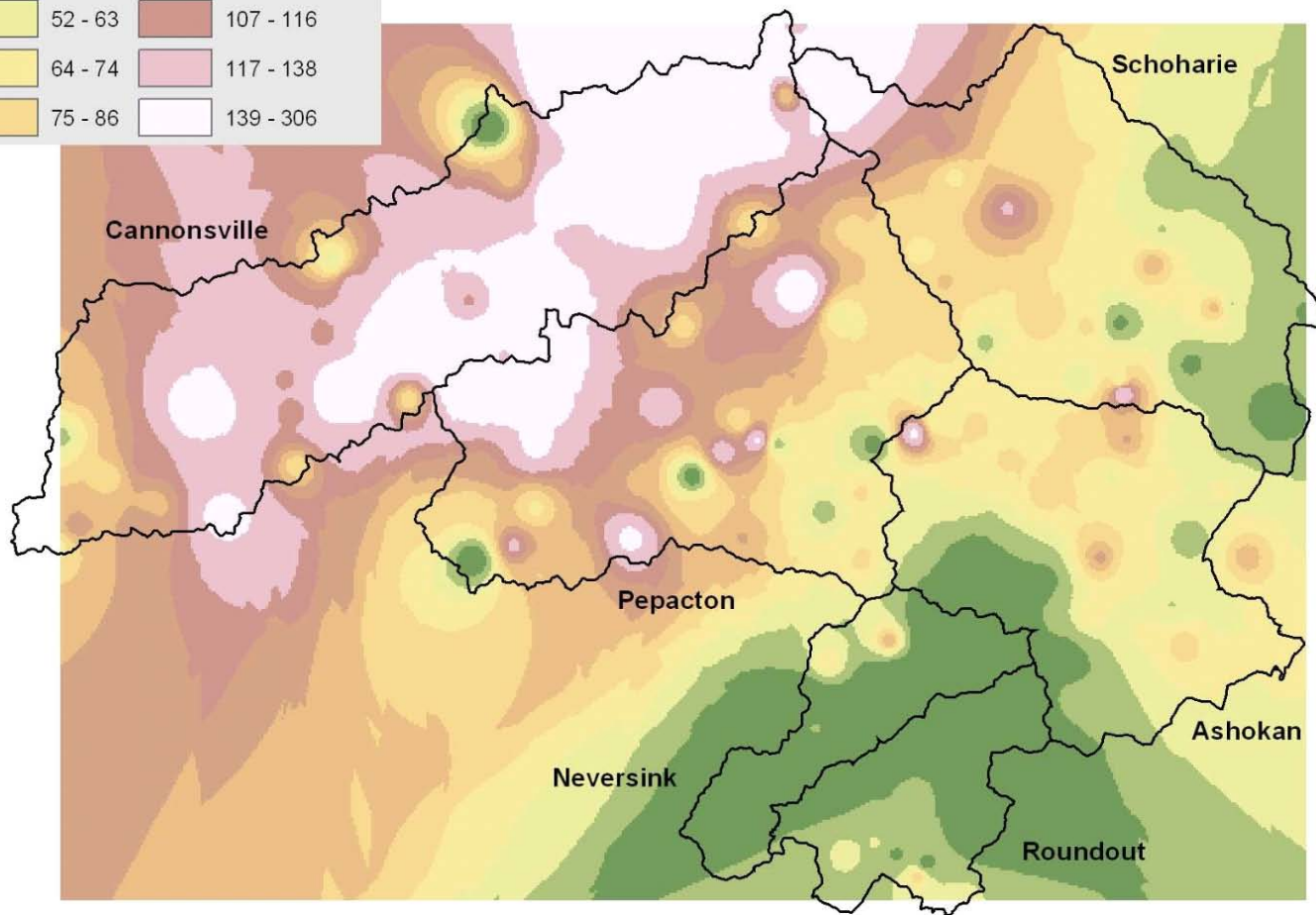
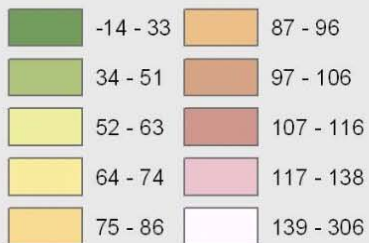
National Atmospheric Deposition Program/National Trends Network
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Surficial Geology

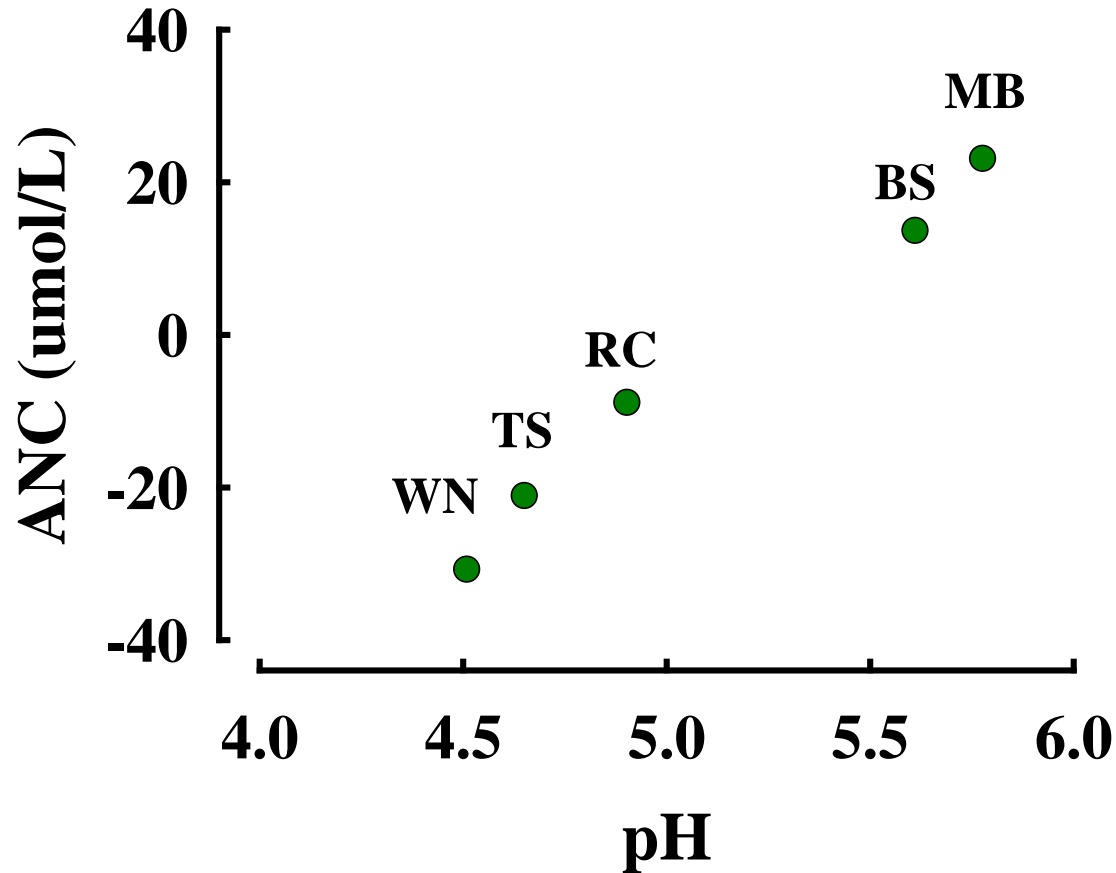


Catskill Stream Survey: ANC (N=180)

Acid Neutralizing Capacity

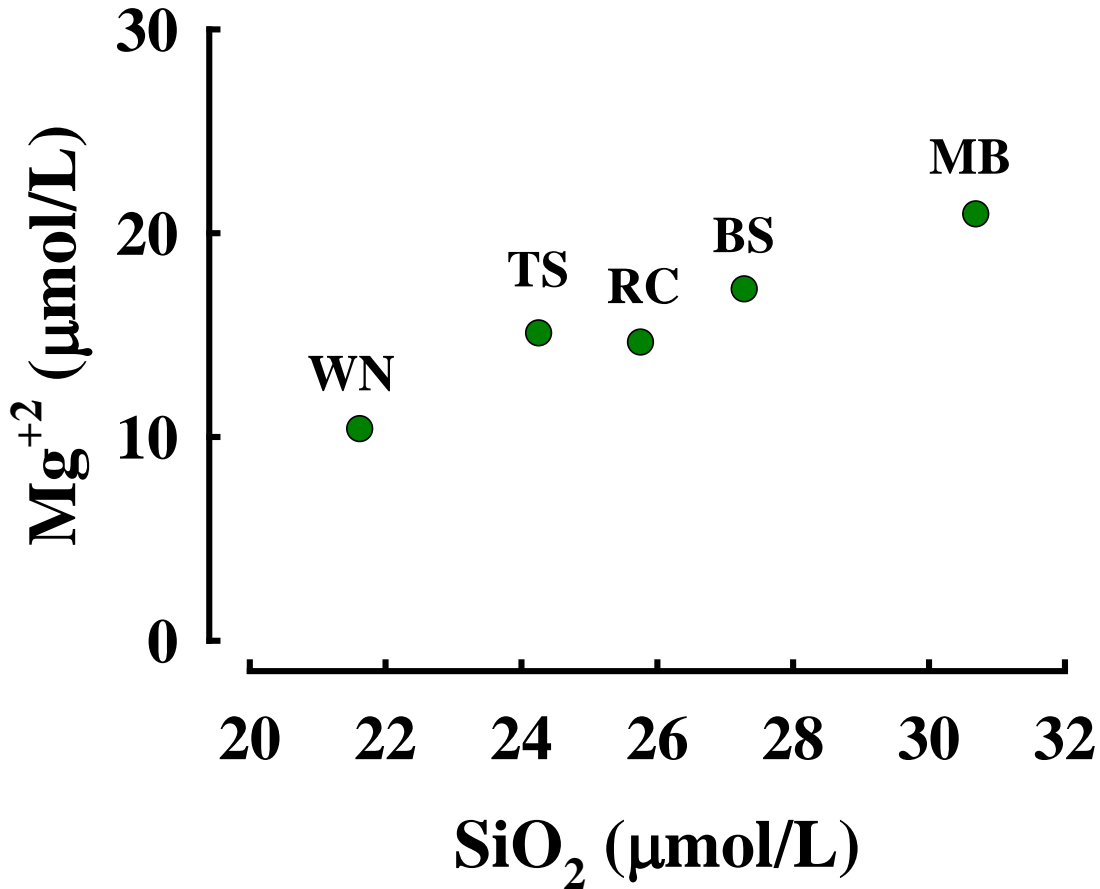


Mean Annual Flow Wt. Concentrations 1991-2008



Most Acidic: **WN**

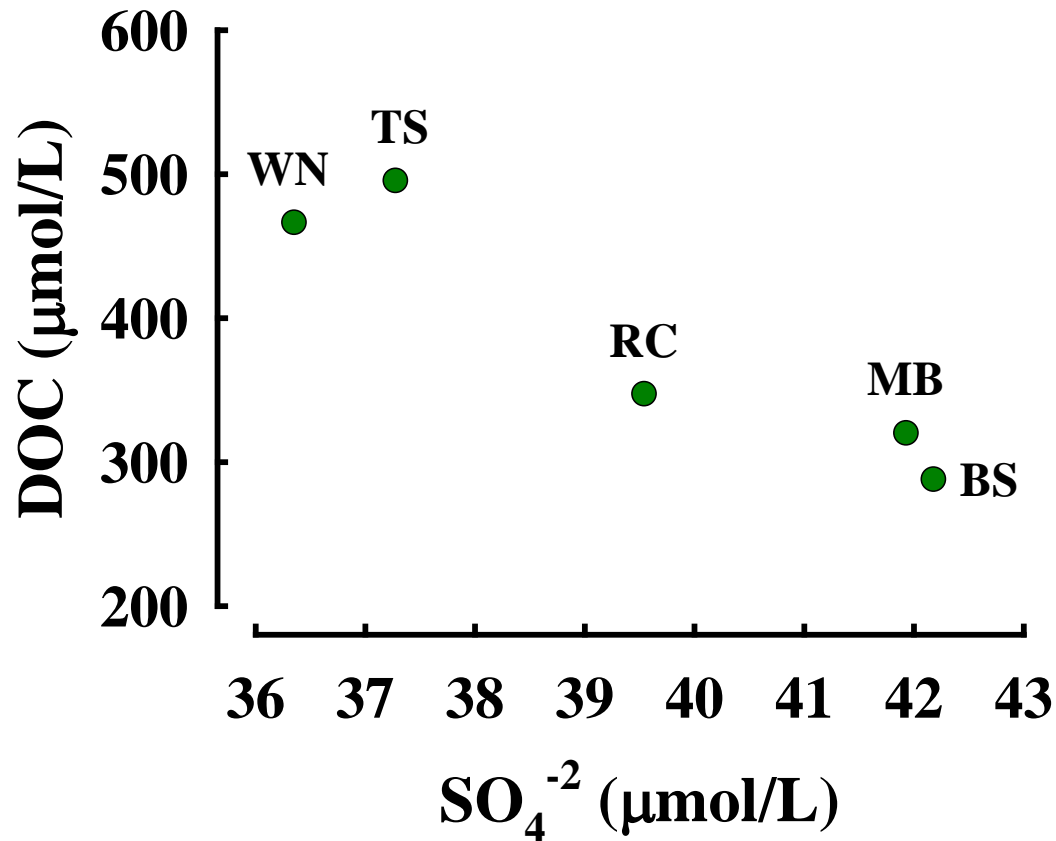
Mean Annual Flow Wt. Concentrations 1991-2008



Highest Silica MB

Highest Mg: **MB**

Mean Annual Flow Wt. Concentrations 1991-2008



Highest Sulfate: **MB & BS**

Highest DOC: **TS & WN**

Catskill LTM Trends

Water Years 1991-2005

Source	pH	SO4	DOC	Ca	Mg	SiO2
Deposition	0.02	-1.15	NA	NS	NS	NA
O-horizon	-0.01	-2.56	34.5	NS	-0.48	NS
Upper B-horizon	-0.01	-2.04	NS	-0.21	-0.33	NS
Lower B-horizon	NS	-2.21	NS	0.30	NS	NS
WN Low Flow	0.01	-1.52	1.91	NS	-0.46	-0.45
BS Low Flow	NS	-2.57	2.99	NS	-0.61	-0.50
RC Low Flow	-0.03	-1.95	5.07	NS	-0.78	-0.82
TS Low Flow	NS	-1.83	3.18	NS	-0.59	-0.41
MB Low Flow	NS	-1.80	0.53	NS	NS	-0.58
WN Stream Water	0.001	-2.65	1.14	NS	-0.38	-0.43
BS Stream Water	NS	-2.42	3.37	NS	-0.52	-0.37
RC Stream Water	-0.02	-1.97	NS	NS	-0.49	-0.41
TS Stream Water	0.01	-1.97	2.14	-0.70	-0.60	-0.33
MB Stream Water	-0.02	-1.97	NS	NS	-0.49	-0.41

All concentrations are in microequivalents per liter per year

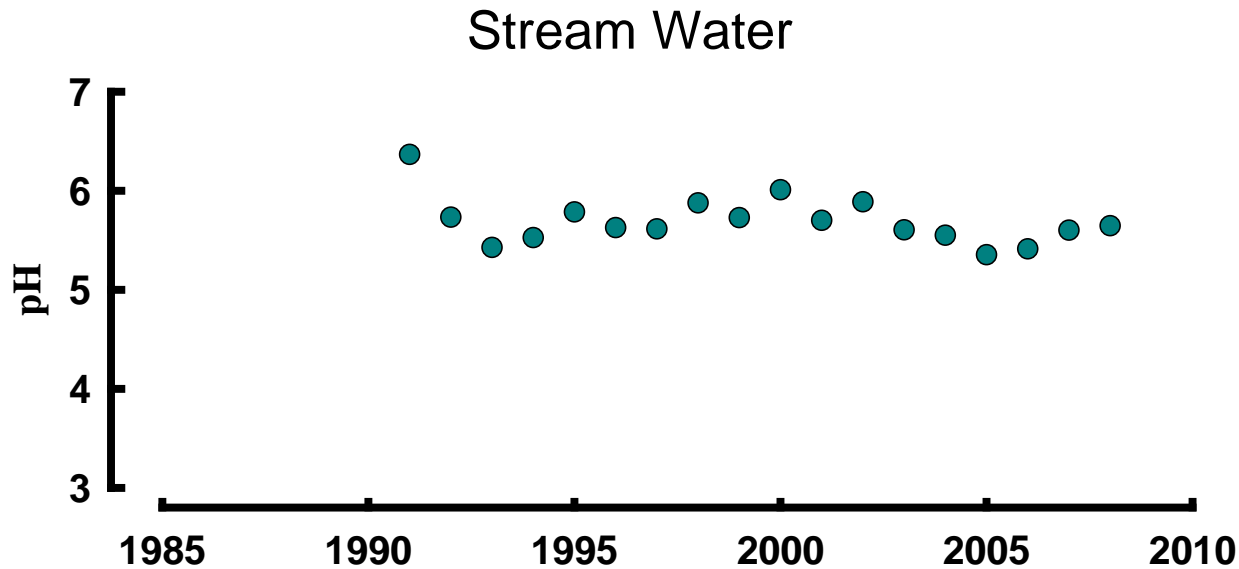
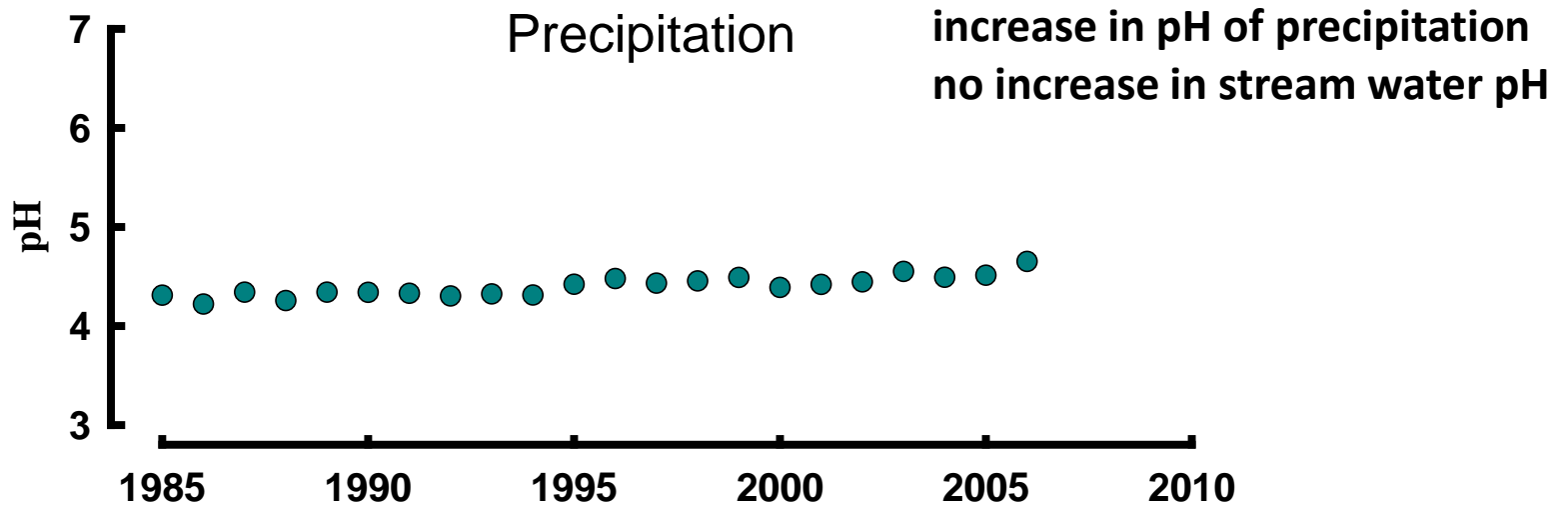
Trends

WY1991-2008

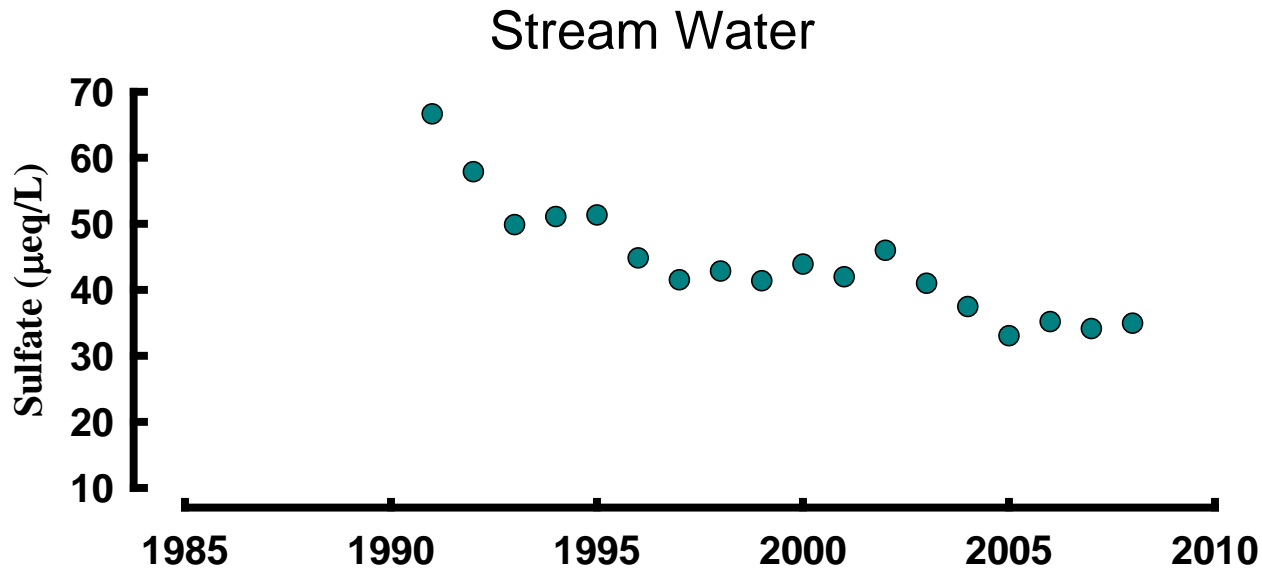
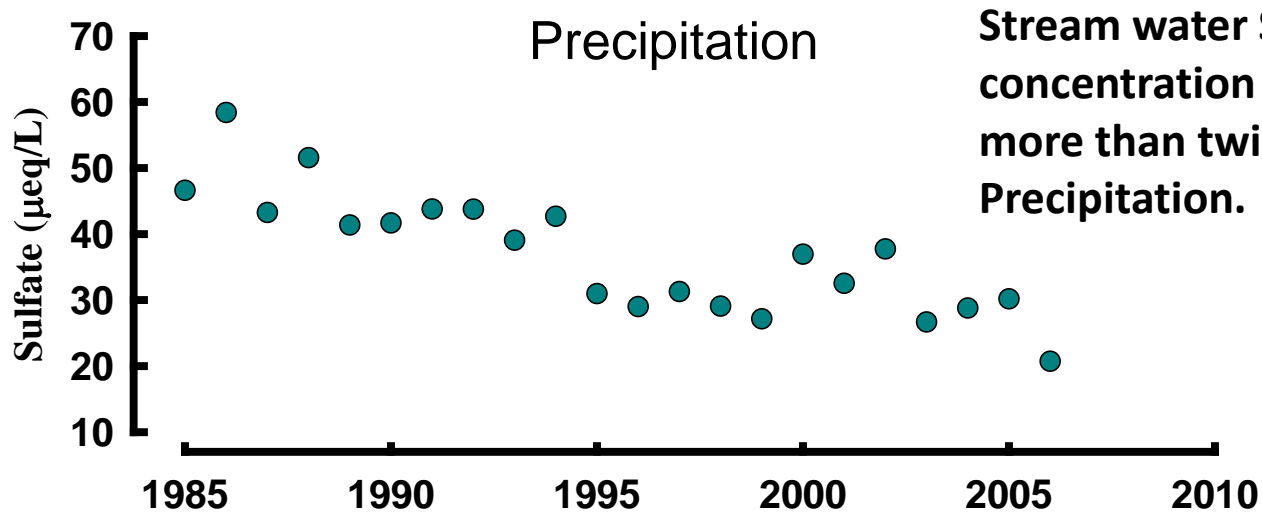
Source	pH	SO4	NO3	DOC	Ca	Mg	SiO2	Almono	Alorg	Altot
BS NADP	0.01	-1.07	-0.51	NA	NS	NS	NA	NA	NA	NA
Winnisook	0.01	-1.92	NS	1.82	-0.32	-0.47	-0.40	-0.20	-0.06	-0.30
Biscuit Brook	NS	-2.42	NS	2.66	-0.49	-0.46	-0.30	-0.05	-0.04	-0.03
Rondout Creek	NS	-2.01	NS	2.20	-0.52	-0.57	-0.40	-0.10	-0.04	-0.10
Tison's Creek	0.02	-1.85	0.21	NS	-0.55	-0.56	-0.30	-0.10	-0.05	-0.20
Main Branch	NS	-2.03	0.29	NS	NS	-0.44	-0.40	-0.04	-0.02	NA

All concentrations are in microequivalents per liter per year

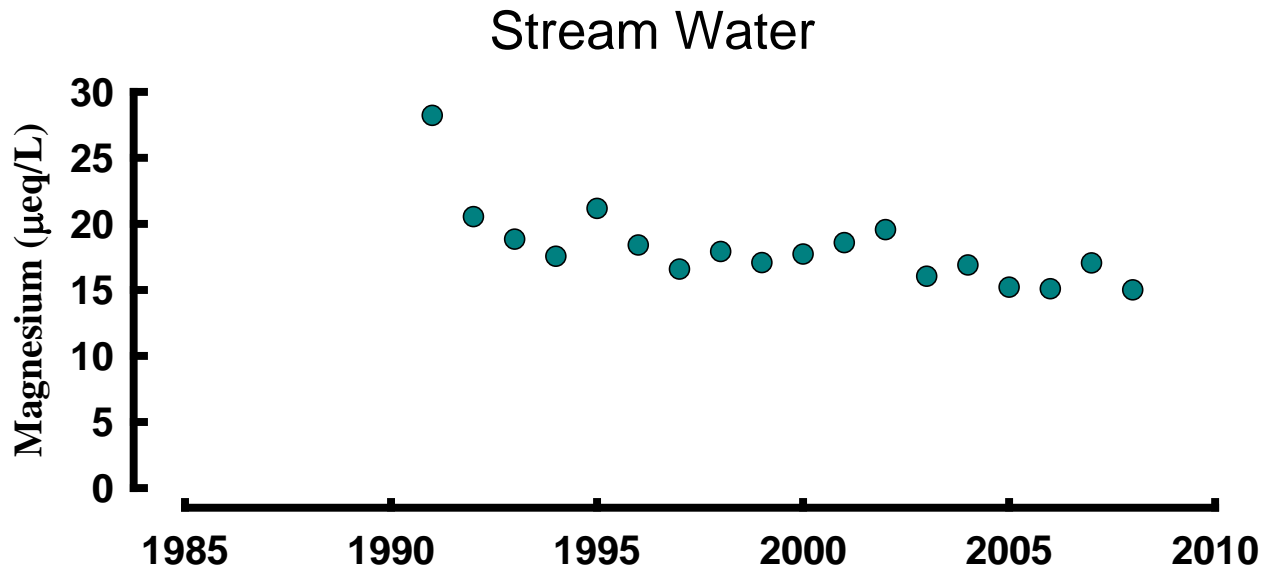
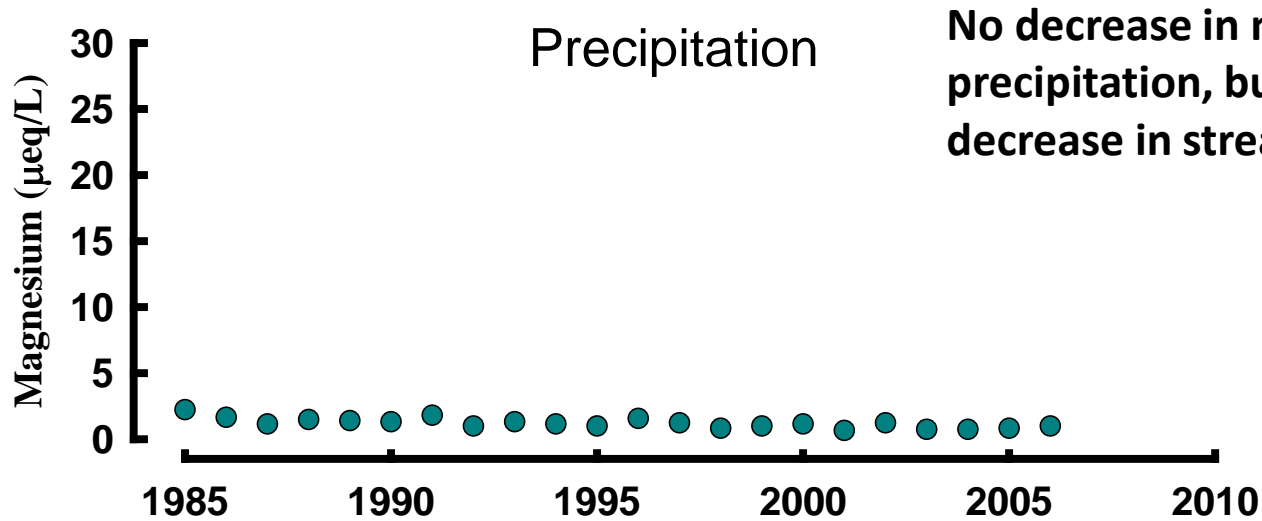
Biscuit Brook Annual Means



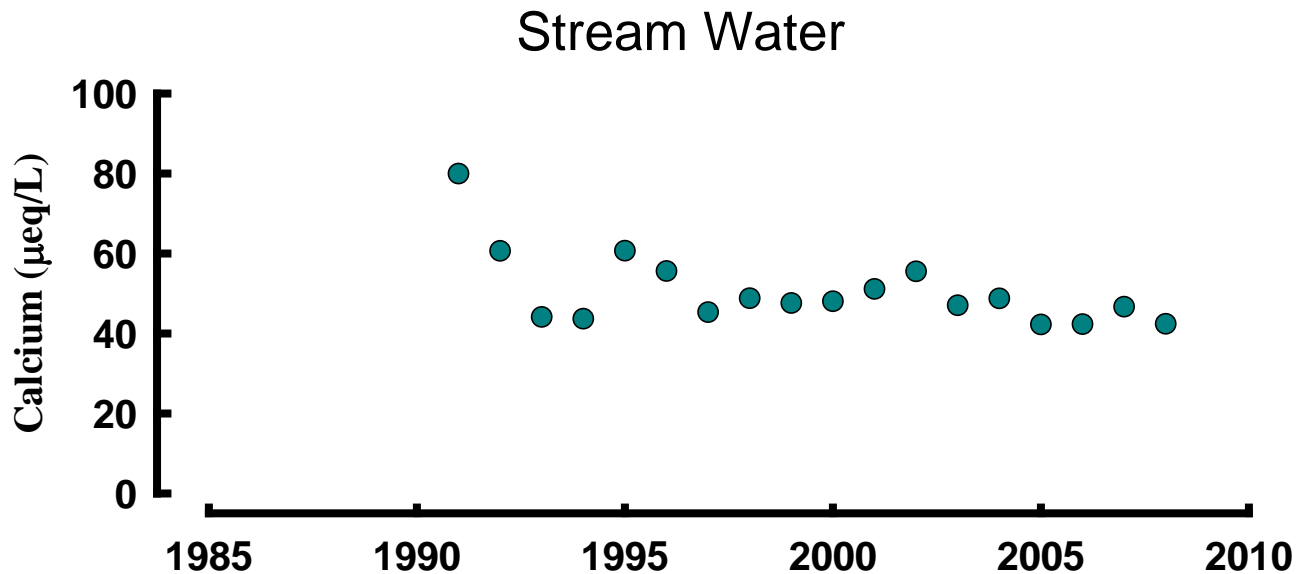
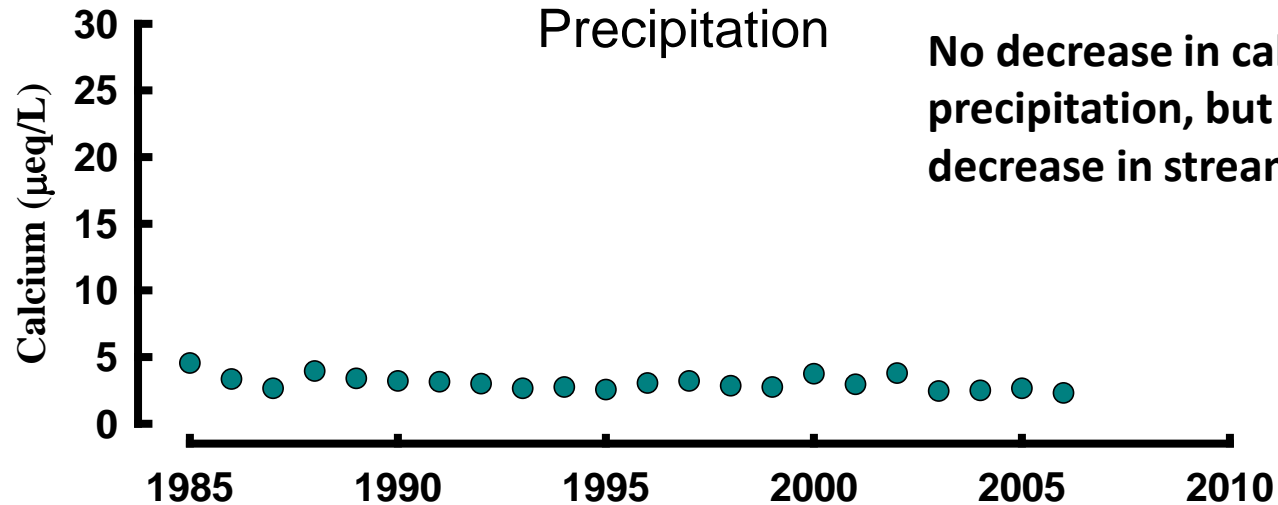
Biscuit Brook Annual Means



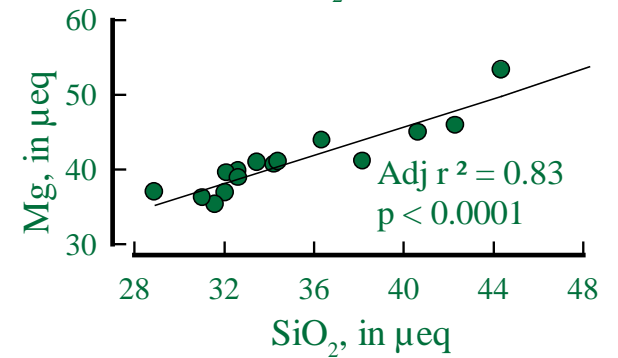
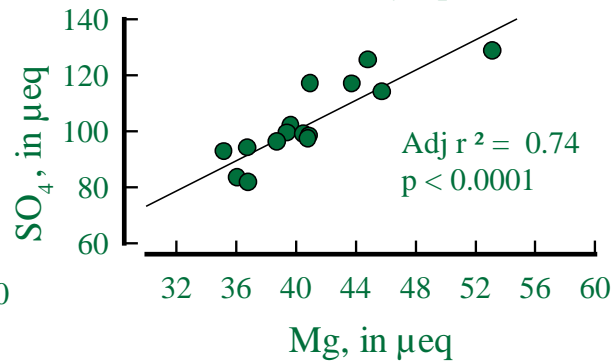
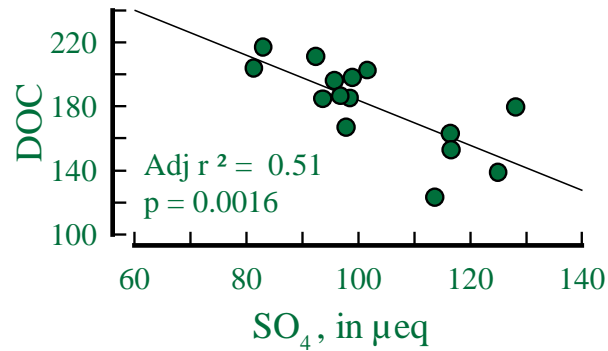
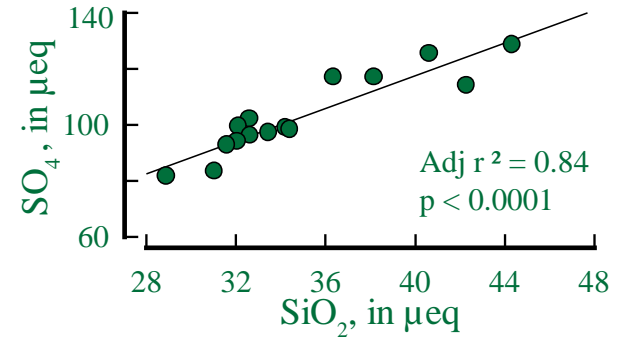
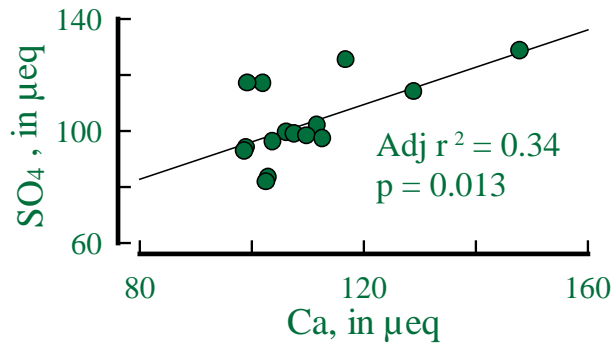
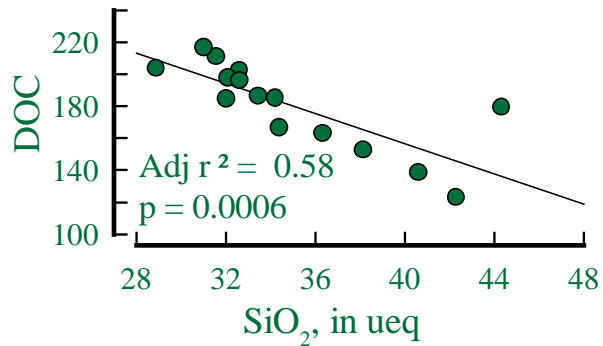
Biscuit Brook Annual Means



Biscuit Brook Annual Means



Relations between mean annual stream water concentrations for Biscuit Brook



pH

- ◆ pH is influenced by many factors:
 - increasing and decreasing trends in stream water at different sites
 - mixing processes in stream water (soil, groundwater and precipitation)
 - regression analyses: no one solute accounted for more than ~40% of variability in mean annual pH

Sulfate

- ◆ The sulfate trends in stream water concentration were double those calculated for deposition
 - evapotranspiration is likely responsible for concentration effect
 - adsorption by mineral and organic soil particles
 - biotic cycling: accumulation of SO_4 in litter and soil organic matter

The decrease in sulfate deposition (reduced anions in solution) leads to a decline in base cations (charge balance).

Calcium

- ◆ There were almost no significant trends, except in soil water

DOC

- There were increasing trends at almost all of the sites, even during low flow.
- The strongest increase was in O-horizon soil water
- There is a significant inverse relationship between mean annual SO_4 and DOC:
 - The ionic strength of soil and stream water decreases due to the decline in sulfate deposition, so more DOC is drawn into solution
- The increase in regional air temperature may be causing a greater rate in decomposition and higher DOC concentrations

Magnesium

- There are consistent decreasing trends in stream water
- Mg seems to balance the drop in anions in stream and soil water rather than Ca

Is this an indication of decreasing bedrock weathering as result of decline in acid deposition?

(supported by mineral composition of Catskills)

quartz (SiO_2)

muscovite ($\text{KAl}_2(\text{AlSi}_3\text{O}_{10})(\text{OH})_2$)

biotite ($\text{K}(\text{Mg,Fe})_3\text{AlSi}_3\text{O}_{10}(\text{OH})_2$)

chlorite ($(\text{Mg,Fe,Al})_{5-6}(\text{Al,Si})_4\text{O}_{10}(\text{OH})_8$)

pyrite (FeS_2)

hornblende ($\text{Ca}_2(\text{Mg,Fe})_4(\text{Al,Fe})\text{Si}_7\text{AlO}_{22}(\text{OH})_2$)

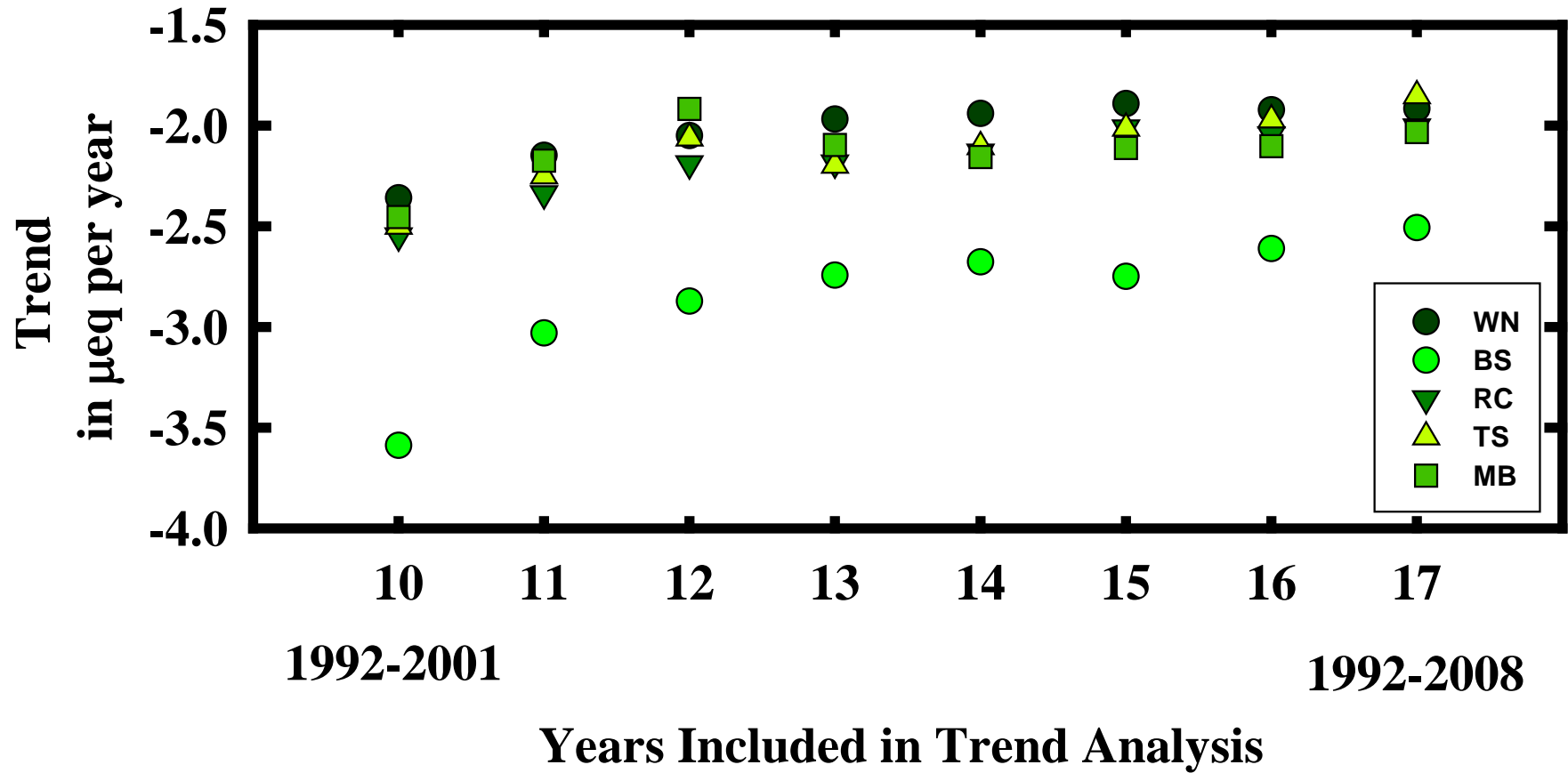
and clays

(Way 1972)

Silica

- There are decreasing trends in SiO_2 in stream water and during low flow
- The strong relationship between Magnesium and Silica supports hypothesis of decline in weathering rate because of decrease in SO_4 deposition
- There are large mineralogical source of silica in these watersheds.
- A decrease in the weathering rate could significantly delay ecosystem recovery from acid deposition

Trend of Sulfate Trends



Trend of DOC Trends

